

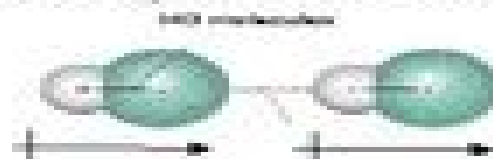
Liquids, Solids & IMF: Practice Assessment

Name _____

Skill - Student will define & differentiate b/w intra- & intermolecular forces, given a diagram.

1. What is the difference between **INTER**molecular forces and **INTRA**molecular forces?

2. Based on the images below, identify the intermolecular forces that are shown by the dotted lines.



3. Using your knowledge of molecular structure, identify the main intermolecular force found between the following substances. You may find it useful to draw Lewis structures to find your answer.

a) PF3 _____

b) Ne _____

c) HF _____

4. Explain how each of the following intermolecular forces, **are formed** between molecules.

London dispersion:

Dipole-dipole:

Hydrogen "bonding":

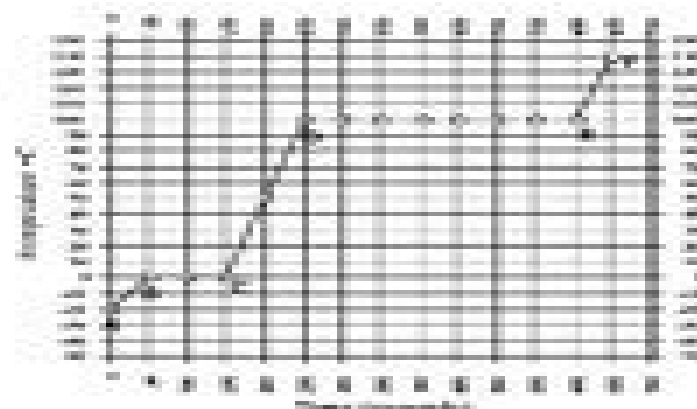
Skill - Student will explain, using IMF, how energy is being used during a phase change.

5. The heating curve represents the heating of water at a constant rate of heat transfer. Answer the following questions using this heating curve.

a. In what part of the curve would water have a definite shape and volume?

b. In what part of the curve would water have a definite volume but not a definite shape?

c. In what part of the curve would water have no definite shape or volume?



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Rosemary Sherrington



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