

Molecular Composition of Gases

CHECK YOUR KNOWLEDGE

ANSWER Answer the following questions in the space provided.

1. _____ The average speed of a gas molecule is most directly related to the _____.
 - (a) volume of the container
 - (b) pressure of the gas
 - (c) temperature of the gas
 - (d) number of molecules in the gas
2. _____ Which value below has the greatest average molecular mass? Assume gas given is at the same temperature.
 - (a) The noble gas with the smallest atomic number
 - (b) The noble gas with the largest atomic number
 - (c) The noble gas with the square root of its atomic number
 - (d) The noble gas with the square of its atomic number
3. _____ Which gas has the greatest average molecular mass? The following are given:
 - nitrogen oxygen carbon dioxide
 - a. _____ Nitrogen is present. Its volume is 25% compared to the volume of oxygen.
 - b. _____ Nitrogen is present. Masses of N_2 and CO_2 are 100 g and 150 g, respectively.
4. _____ Which will be the sample of an ideal gas at 20°C?
 - a. _____ One mole, molecular gas with the smallest molar mass.
 - b. _____ One mole, molecular gas with the largest molar mass.
 - c. _____ One mole, molecular gas with the largest molar mass.
 - d. _____ One mole, molecular gas with the largest molar mass.

PROBLEM SET Solve the following problems in the space provided.

1. _____ Calculate the volume of 1 mole of an ideal gas at 20°C and 1 atm.

Review Molecular Composition Of Gases Short Answer

LP Steffe



Review Molecular Composition Of Gases Short Answer:

The book delves into Review Molecular Composition Of Gases Short Answer. Review Molecular Composition Of Gases Short Answer is an essential topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Review Molecular Composition Of Gases Short Answer, encompassing both the fundamentals and more intricate discussions.

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