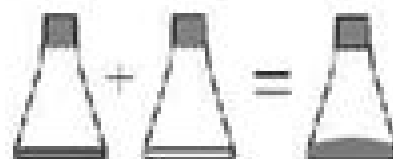


Explain the conservation of mass.

In a chemical reaction, the mass of the final product is the same as the mass of the reacting substances because atoms are rearranged but not destroyed.



Example:

When coal (carbon) is burned in the presence of air (oxygen), another gas (carbon dioxide) is formed.



Mark the statements TRUE or FALSE.

When the coal was burned, the carbon was destroyed.

Atoms are not destroyed, but bond with other molecules to form new substances.

The conservation of mass only applies to solids.

The number of atoms always remains the same.

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