

Three-mechanism reactions

10. A reaction $\text{A} \rightarrow \text{B}$ has $E_a = 100 \text{ kJ mol}^{-1}$. At 25°C , $k = 1.0 \times 10^{-5} \text{ s}^{-1}$. At 35°C , $k = 1.0 \times 10^{-4} \text{ s}^{-1}$.

Which statement is correct for this reaction mechanism?

- a. A and B are tetrahedral species of $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_3$
- b. A and B are tetrahedral species of $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$
- c. A is the tetrahedral tetrahedral species of $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$
- d. A and B are tetrahedral species of $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$

11. The energy profile diagram for the reaction $\text{A} \rightarrow \text{B}$ is shown below. The energy of A is 100 kJ mol^{-1} .

Which statement is correct for this reaction?

- a. The reaction is exothermic.
- b. The reaction is endothermic.
- c. The reaction is exothermic.
- d. The reaction is endothermic.

12. The energy profile diagram for the reaction $\text{A} \rightarrow \text{B}$ is shown below. The energy of A is 100 kJ mol^{-1} .

13. The reaction is exothermic. The energy of A is 100 kJ mol^{-1} . The energy of B is 50 kJ mol^{-1} .

14. The reaction is exothermic. The energy of A is 100 kJ mol^{-1} . The energy of B is 50 kJ mol^{-1} .

15. The reaction is exothermic. The energy of A is 100 kJ mol^{-1} . The energy of B is 50 kJ mol^{-1} .

16. The reaction is exothermic. The energy of A is 100 kJ mol^{-1} . The energy of B is 50 kJ mol^{-1} .

- a. 100 kJ mol^{-1}
- b. 50 kJ mol^{-1}
- c. 0 kJ mol^{-1}

17. The reaction is exothermic. The energy of A is 100 kJ mol^{-1} . The energy of B is 50 kJ mol^{-1} .

18. The reaction is exothermic. The energy of A is 100 kJ mol^{-1} . The energy of B is 50 kJ mol^{-1} .

19. The reaction is exothermic. The energy of A is 100 kJ mol^{-1} . The energy of B is 50 kJ mol^{-1} .

- a. 100 kJ mol^{-1}
- b. 50 kJ mol^{-1}
- c. 0 kJ mol^{-1}
- d. 100 kJ mol^{-1}

20. The reaction is exothermic. The energy of A is 100 kJ mol^{-1} . The energy of B is 50 kJ mol^{-1} .

21. The reaction is exothermic. The energy of A is 100 kJ mol^{-1} . The energy of B is 50 kJ mol^{-1} .

22. The reaction is exothermic. The energy of A is 100 kJ mol^{-1} . The energy of B is 50 kJ mol^{-1} .

23. The reaction is exothermic. The energy of A is 100 kJ mol^{-1} . The energy of B is 50 kJ mol^{-1} .

24. The reaction is exothermic. The energy of A is 100 kJ mol^{-1} . The energy of B is 50 kJ mol^{-1} .

25. The reaction is exothermic. The energy of A is 100 kJ mol^{-1} . The energy of B is 50 kJ mol^{-1} .

- a. 100 kJ mol^{-1}
- b. 50 kJ mol^{-1}
- c. 0 kJ mol^{-1}
- d. 100 kJ mol^{-1}

26. The reaction is exothermic. The energy of A is 100 kJ mol^{-1} . The energy of B is 50 kJ mol^{-1} .

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Ying-Ying Zheng



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