

Integrated Physical Science – High School

Standard assignment 21 schedule

Chapter 18: Materials

- Key Terms
- Review Questions: 1, 3, 4, 7, 10, 11
- Questions: 1, 3, 10

Chapter 19: Atoms and Subatomic

Key Terms and Matching Questions

- 1. atomic spectrum
- 2. continuous model
- 3. physical model
- 4. quantum spectrum model
- 5. quantum
- 6. quantum hypothesis
- 7. shell
- 8. spectroscopy
- 9. hydrogen spectrum

Review Questions

1. Why are atoms invisible to visible light?
Atoms are smaller than the wavelengths of visible light. Therefore, they cannot be seen.
2. What is the function of an atomic model?
An atomic model predicts the behavior of a system that we cannot see.
3. What occurs as atoms absorb light?
Atoms gain all light as they are subjected to energy.
4. Why do we say spectral lines are like fingerprints of the elements?
The pattern of each element will only reflect the quantum of light.

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