

Chemistry Semester 2 Final Review Topics

Understand the definitions, identification, notation, characteristics, and processes relating to these topics.

- General
 - Use of the periodic table in identifying charges, atomic mass, metals or nonmetals, and states of matter.
- Periodic Trends
 - Electronegativity
- Ionic Bonds
 - Naming
 - Formulas
 - Identifying charges and subscripts
 - Dot diagrams
- Covalent Bonds
 - Naming
 - Formulas
 - Identifying prefixes and suffixes
 - Dot diagrams
- Polarity
 - Finding electronegativity difference
 - Partial charges
 - Individual bond dipole moments
 - Overall molecular dipole moment
 - Polar vs. Nonpolar characteristics
 - Molecular Geometry
- Skeleton Equations
 - Going from words to equation or equation to words
 - Identifying reactants and products
 - Identifying states of matter
 - Writing chemical formulas properly
 - Identifying when heat or enzymes are added
- Balancing chemical equations
 - Understanding how to do the math of balancing
 - Understanding how to write, interpret, and use the coefficients
- Moles Conversions
 - Mass to Mole
 - Particles to Mole
 - Volume to Mole
 - Substance to another Substance using the balanced chemical equation
- Types of chemical reactions
 - Identifying types and predicting products
 - Synthesis
 - Decomposition
 - Single Replacement
 - Double Replacement
 - Combustion

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