

Chemical bonding - answers

1. Atoms form molecules by sharing electrons and those are known as covalently bonded molecules. As same, by donating electrons to non-metals from metals the ionic compounds are forms.

a) Group the following compounds – ionic or covalent.

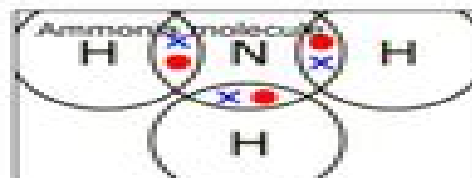
Carbon dioxide Magnesium Aluminium oxide water sulfur trioxide
Potassium iodide methane ethanol oxygen ammonium chloride ammonia gas
Diamond copper nitrate silicon dioxide sodium chloride hydrogen chloride gas

Covalent compounds	Ionic compounds
Carbon dioxide	Aluminium oxide
Water	Potassium iodide
Sulfur trioxide	Copper nitrate
Methane	Sodium chloride
Ethanol	
Oxygen	
Ammonia gas	
Silicon dioxide	
Hydrogen gas	

b). Simple molecular compounds can be formed by sharing electrons only. Give 3 examples for simple molecular compounds.

1.....methanol 2.....sulfur dioxide 3.....hydrogen sulfide gas (any covalent compound)

c. Draw dot cross diagrams to show the bonding in: i) hydrogen chloride molecule ii) ammonia molecule



d) Explain why the bond in a nitrogen molecule is called a triple bond.

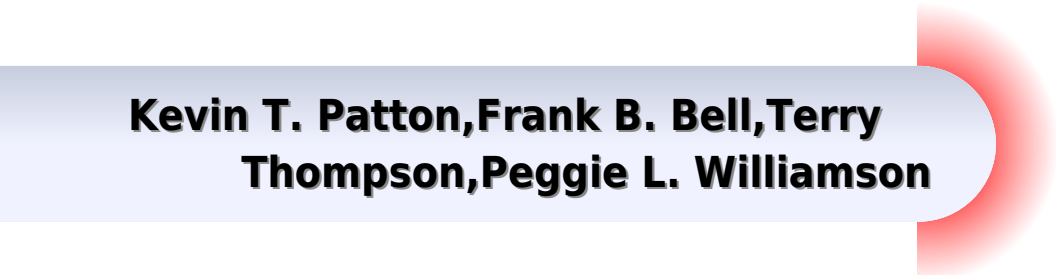
Two nitrogen atoms share three electrons from each atom to form three covalent bonds.

e). How many electrons are need to form a covalent bond?

Two electrons / a pair of electrons

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**Kevin T. Patton, Frank B. Bell, Terry
Thompson, Peggie L. Williamson**



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