

CHAPTER 6 REVIEW*Chemical Bonding***MIXED REVIEW**

SHORT ANSWER Answer the following questions in the space provided.

1. Name the type of energy that is a measure of strength for each of the following types of bonds:

_____ lattice energy _____	a. ionic bond
_____ bond energy _____	b. covalent bond
_____ enthalpy of vaporization _____	c. metallic bond

2. Use the electronegativity values shown in **Figure 20**, on page 161 of the text, to determine whether each of the following bonds is nonpolar covalent, polar covalent, or ionic.

_____ ionic _____	a. H—F	_____ nonpolar covalent _____	d. H—H
_____ ionic _____	b. Na—Cl	_____ polar covalent _____	e. H—C
_____ polar covalent _____	c. H—O	_____ polar covalent _____	f. H—N

3. How is a hydrogen bond different from an ionic or covalent bond?

A hydrogen bond is a dipole-dipole attraction between a partially positive hydrogen
atom and the unshared electron pair of a strongly electronegative atom such as O,
N, or F. Unlike ionic or covalent bonds, in which electrons are given up or shared,
the hydrogen bond is a weaker attraction. Hydrogen bonds are generally
intermolecular, while ionic and covalent bonds occur between ions or atoms
respectively.

4. H₂S and H₂O have similar structures and their central atoms belong to the same group. Yet H₂S is a gas at room temperature and H₂O is a liquid. Use bonding principles to explain why this is.

Oxygen has higher electronegativity than sulfur, which creates a highly polar bond.
Increased polarity in H₂O bonds means a stronger intermolecular attraction, making
water a liquid at room temperature. Hydrogen bonding exists between water
molecules, but not between hydrogen sulfide molecules.

Review Chemical Bonding Answer Key

Guillaume Favre

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Review Chemical Bonding Answer Key:

Whispering the Techniques of Language: An Mental Journey through **Review Chemical Bonding Answer Key**

In a digitally-driven earth where displays reign great and immediate communication drowns out the subtleties of language, the profound strategies and mental subtleties hidden within phrases frequently get unheard. Yet, located within the pages of **Review Chemical Bonding Answer Key** a interesting fictional value sporting with raw thoughts, lies an exceptional journey waiting to be undertaken. Penned by an experienced wordsmith, that wonderful opus attracts visitors on an introspective trip, delicately unraveling the veiled truths and profound impact resonating within the fabric of every word. Within the mental depths of the emotional review, we can embark upon a honest exploration of the book is key subjects, dissect their fascinating writing model, and succumb to the effective resonance it evokes strong within the recesses of readers hearts.

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