

Reversible Reactions & Chemical Equilibrium

- **Reversible reaction** = chemical reaction that can occur in either direction.
- Forward reaction: $A \rightarrow B$
- Reverse reaction: $A \leftarrow B$
- Usually combined: $A \rightleftharpoons B$
- **Chemical equilibrium** = point at which the rate of forward reaction **equals** the rate of reverse reaction.

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