

Answer following



- 1. Substrate and temperature: 20°C $100\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$
- 2. A good substrate: $100\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$ $100\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$
- 3. A poor substrate: $10\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$ $10\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$

Calculate all values of K_m and V_{max} for the following reactions:
 (Note: K_m is the substrate concentration at which the reaction rate is half of V_{max})

- 1. $K_m = 10\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$ $V_{max} = 100\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$
- 2. $K_m = 10\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$ $V_{max} = 100\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$
- 3. $K_m = 10\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$ $V_{max} = 100\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$
- 4. $K_m = 10\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$ $V_{max} = 100\text{ mg} \cdot \text{min}^{-1} \cdot \text{mg}^{-1}$



Exercise 10

Graph the following data, and determine the K_m and V_{max} for the reaction.



Graph the following data, and determine the K_m and V_{max} for the reaction.



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