

Macromolecules - Answers

1. An amino acid contains a central carbon atom, a hydrogen atom, a carboxyl group, an amino group group, and a differing R group.
2. What is a dipeptide? What type of bond is involved? = two amino acids joined via a peptide bond.
3. A nucleotide is a monomer of nucleic acids. There are 4 (how many) different types of nucleotides. (for either RNA or DNA)
4. State two functions of fats.
long term energy storage
insulation
lubrication
cushioning
5. What is the structure of a phospholipid? hydrophilic head group, with a phosphate - always charged and a group of atoms (ex. choline, serine); two hydrophobic tails, fatty acids with long hydrocarbon chains
6. ATP is a molecule that stores a great deal of energy within its phosphate bonds. True or False? True
7. Name and state the structure of a common dissaccharide.
two sugars joined via covalent bond; ex. glucose-glucose = maltose; glucose-fructose = sucrose
8. Name and state the structure of a common polysaccharide.
glycogen and starch = polymers of alpha glucose; storage form of glucose in animals and plants
cellulose = polymers of beta glucose; rigid structure in plant cell walls
chitin = polymer of glucosamine; very strong and flexible; found in exoskeletons of arthropods
9. Heating an enzyme to a certain temperature usually destroys its enzymatic activity even though the same sequence of amino acids remains. Name and explain the levels of structure that have been affected by heating.
secondary structure = alpha helices, beta pleated sheets, random coils
tertiary structure = overall three dimensional conformation, or shape, of the polypeptide
quaternary structure = assembly of subunits

primary structure = chain of amino acids; usually remains intact upon denaturation due to strong peptide bonds.

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Wenbin Ji



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Unveiling the Energy of Verbal Artistry: An Emotional Sojourn through **Prentice Hall Biology Study Guide Macromolecules Answers**

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Table of Contents Prentice Hall Biology Study Guide Macromolecules Answers

1. Understanding the eBook Prentice Hall Biology Study Guide Macromolecules Answers
 - The Rise of Digital Reading Prentice Hall Biology Study Guide Macromolecules Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Prentice Hall Biology Study Guide Macromolecules Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Prentice Hall Biology Study Guide Macromolecules Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Prentice Hall Biology Study Guide Macromolecules Answers

- Personalized Recommendations
- Prentice Hall Biology Study Guide Macromolecules Answers User Reviews and Ratings
- Prentice Hall Biology Study Guide Macromolecules Answers and Bestseller Lists
- 5. Accessing Prentice Hall Biology Study Guide Macromolecules Answers Free and Paid eBooks
 - Prentice Hall Biology Study Guide Macromolecules Answers Public Domain eBooks
 - Prentice Hall Biology Study Guide Macromolecules Answers eBook Subscription Services
 - Prentice Hall Biology Study Guide Macromolecules Answers Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
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 - Prentice Hall Biology Study Guide Macromolecules Answers Enhanced eBook Features
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 - Highlighting and Note-Taking Prentice Hall Biology Study Guide Macromolecules Answers
 - Interactive Elements Prentice Hall Biology Study Guide Macromolecules Answers
- 8. Staying Engaged with Prentice Hall Biology Study Guide Macromolecules Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Prentice Hall Biology Study Guide Macromolecules Answers
- 9. Balancing eBooks and Physical Books Prentice Hall Biology Study Guide Macromolecules Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Prentice Hall Biology Study Guide Macromolecules Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Prentice Hall Biology Study Guide Macromolecules Answers
 - Setting Reading Goals Prentice Hall Biology Study Guide Macromolecules Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Prentice Hall Biology Study Guide Macromolecules Answers

- Fact-Checking eBook Content of Prentice Hall Biology Study Guide Macromolecules Answers
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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