



Name \_\_\_\_\_ Class \_\_\_\_\_ Date \_\_\_\_\_

**1**

How many atoms of **oxygen (O)** are in the **chemical formula** below?

- A** 4
- B** 5
- C** 7
- D** 12

**3**

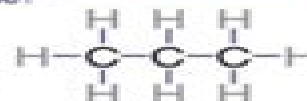
Carbon (C) and hydrogen (H) create **covalent bonds** to share **electrons** between them. What is the total number of electrons being shared between **carbon** and **hydrogen** in the molecule of propane shown below?

- A** 8
- B** 11
- C** 16
- D** 32

**2**

The diagram shown below is a **structural formula** for the gas **propane**. What would be the correct **chemical formula** for propane?

- A**  $\text{C}_3\text{H}_6$
- B**  $\text{CH}_6$
- C**  $8\text{H}_3\text{C}$
- D**  $\text{CH}$

**4**

The **reason** why the bonds between carbon (C) and hydrogen (H) are **covalent** is \_\_\_\_\_

- A** C and H do not have electrons to give away
- B** C and H are both nonmetals
- C** C and H are both metalloids
- D** C and H can bond only by sharing electrons

**5**

The diagram below represents the **structural formula** for several **water molecules**. How many **molecules** are represented?

- A** 3
- B** 5
- C** 10
- D** 15

**6**

Explain the **arrangement** of the molecules in the diagram of water below.

- A** it is a lucky arrangement
- B** water molecules are sticky
- C** water molecules have polarity
- D** water molecules are magnetic

**7**

Water molecules have some **polarity** in their structure. Using the diagram below, notice how the ends of H atoms (**positive**) and O atoms (**negative**) meet. This property is the **reason** that \_\_\_\_\_

- A** water forms droplets easily
- B** water boils easily
- C** ice melts easily
- D** water has color

**8**

Since H atoms are slightly **positive** and the O atoms are slightly **negative**, molecules of water have **polarity**. This makes water a **good** \_\_\_\_\_

- A** solute
- B** refrigerant
- C** solvent
- D** magnet

**9**

The chemical formula for the **oxygen** that we breathe is \_\_\_\_\_

- A** O
- B**  $\text{O}_2$
- C**  $\text{O}_3$
- D**  $2\text{O}$

**10**

How many **molecules of oxygen** are represented in the chemical formula below?

- A** 2
- B** 4
- C** 6
- D** 18



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### **Table of Contents Oaa 8th Grade Science Practice Questions**

1. Understanding the eBook Oaa 8th Grade Science Practice Questions
  - The Rise of Digital Reading Oaa 8th Grade Science Practice Questions
  - Advantages of eBooks Over Traditional Books
2. Identifying Oaa 8th Grade Science Practice Questions
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Oaa 8th Grade Science Practice Questions
  - User-Friendly Interface
4. Exploring eBook Recommendations from Oaa 8th Grade Science Practice Questions
  - Personalized Recommendations

- Oaa 8th Grade Science Practice Questions User Reviews and Ratings
- Oaa 8th Grade Science Practice Questions and Bestseller Lists
- 5. Accessing Oaa 8th Grade Science Practice Questions Free and Paid eBooks
  - Oaa 8th Grade Science Practice Questions Public Domain eBooks
  - Oaa 8th Grade Science Practice Questions eBook Subscription Services
  - Oaa 8th Grade Science Practice Questions Budget-Friendly Options
- 6. Navigating Oaa 8th Grade Science Practice Questions eBook Formats
  - ePub, PDF, MOBI, and More
  - Oaa 8th Grade Science Practice Questions Compatibility with Devices
  - Oaa 8th Grade Science Practice Questions Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Oaa 8th Grade Science Practice Questions
  - Highlighting and Note-Taking Oaa 8th Grade Science Practice Questions
  - Interactive Elements Oaa 8th Grade Science Practice Questions
- 8. Staying Engaged with Oaa 8th Grade Science Practice Questions
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Oaa 8th Grade Science Practice Questions
- 9. Balancing eBooks and Physical Books Oaa 8th Grade Science Practice Questions
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Oaa 8th Grade Science Practice Questions
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Oaa 8th Grade Science Practice Questions
  - Setting Reading Goals Oaa 8th Grade Science Practice Questions
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Oaa 8th Grade Science Practice Questions
  - Fact-Checking eBook Content of Oaa 8th Grade Science Practice Questions

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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