

## 9<sup>th</sup> Grade Physical Science Chemical Bonding Study Guide

### Chemical Bond Descriptions

#### Ionic Bond

Bond between a metal and nonmetal  
Transfers electrons  
Has a rigid crystal structure

#### Covalent Bond

Shares electrons equally

#### Metallic Bond

Bond between a metal and a metal  
That forms metal alloys  
Allows electrons to flow freely  
Conducts electricity at all times

#### Polar Covalent Bond

Unequal sharing of electrons  
Has a slightly positive charge on one end

A chemical formula has symbols and subscripts that tell the type of ratio of atoms in a compound.

The chemical formula for glucose is  $C_{12}H_{22}O_{11}$ . How many elements are in the compound? 3: C is an element, H is an element, and O is an element.

The chemical formula for glucose is  $C_{12}H_{22}O_{11}$ . How many atoms are in the compound? 45: You add the number of atoms.  $12+22+11=45$

The chemical formula for glucose is  $C_{12}H_{22}O_{11}$ . What is the ratio of atoms in the compound? 12:22:11

The properties of a compound depend on the elements and how they are arranged.

The interaction of electrons between elements determines the kind of bond that is formed.

The properties of elements in compounds are usually different from the properties of the elements alone.

The smell of a compound is usually determined by the shape of the molecules that make them.

The chemical formula for a compound having one barium ion and two chlorine ions is  $BaCl_2$ .

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