

QUESTION 1

- 1.1. C ✓✓
- 1.2. D ✓✓
- 1.3. B ✓✓
- 1.4. A ✓✓
- 1.5. A ✓✓

[10]

QUESTION 2

2.1.
2.1.1. A pure substance is a substance that cannot be separated into simpler components by physical methods. ✓✓

- 2.1.2.1 Element ✓
- 2.1.2.2 Consists of only one type of atom. ✓
- 2.1.2.3 It is a combination of different metals (copper and zinc). ✓

2.2. Pots and pans are made of metal, because metal is a good conductor which allows heat to be transferred so that food can cook. ✓ The handles are insulators so that you do not burn your hands when you pick up a hot pot. ✓

- 2.3.1. (a) Temperature ✓
(b) Time ✓
- 2.3.2. Boiling point is the temperature of a liquid at which its vapour pressure equals to the external (atmospheric) pressure. ✓
- 2.3.3. The temperature remains constant because the energy is absorbed by the molecules. ✓
The strength of the intermolecular forces between the molecules decreases. ✓
The particles move faster and spaces between the particles increases. ✓
The substance changes phase from liquid to gas. ✓

[15]

Physical Sciences Chemistry P2 Half Yearly Examination

2013 Memorandum

Peipei Pang

Physical Sciences Chemistry P2 Half Yearly Examination 2013 Memorandum:

The book delves into Physical Sciences Chemistry P2 Half Yearly Examination 2013 Memorandum. Physical Sciences Chemistry P2 Half Yearly Examination 2013 Memorandum is an essential topic that must be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Physical Sciences Chemistry P2 Half Yearly Examination 2013 Memorandum, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Physical Sciences Chemistry P2 Half Yearly Examination 2013 Memorandum
- Chapter 2: Essential Elements of Physical Sciences Chemistry P2 Half Yearly Examination 2013 Memorandum
- Chapter 3: Physical Sciences Chemistry P2 Half Yearly Examination 2013 Memorandum in Everyday Life
- Chapter 4: Physical Sciences Chemistry P2 Half Yearly Examination 2013 Memorandum in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Physical Sciences Chemistry P2 Half Yearly Examination 2013 Memorandum. The first chapter will explore what Physical Sciences Chemistry P2 Half Yearly Examination 2013 Memorandum is, why Physical Sciences Chemistry P2 Half Yearly Examination 2013 Memorandum is vital, and how to effectively learn about Physical Sciences Chemistry P2 Half Yearly Examination 2013 Memorandum.
3. In chapter 2, the author will delve into the foundational concepts of Physical Sciences Chemistry P2 Half Yearly Examination 2013 Memorandum. The second chapter will elucidate the essential principles that must be understood to grasp Physical Sciences Chemistry P2 Half Yearly Examination 2013 Memorandum in its entirety.
4. In chapter 3, the author will examine the practical applications of Physical Sciences Chemistry P2 Half Yearly Examination 2013 Memorandum in daily life. This chapter will showcase real-world examples of how Physical Sciences Chemistry P2 Half Yearly Examination 2013 Memorandum can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of Physical Sciences Chemistry P2 Half Yearly Examination 2013 Memorandum in specific contexts. This chapter will explore how Physical Sciences Chemistry P2 Half Yearly Examination 2013 Memorandum is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Physical Sciences Chemistry P2 Half Yearly Examination 2013 Memorandum. This chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Physical Sciences Chemistry P2 Half Yearly Examination 2013 Memorandum.

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