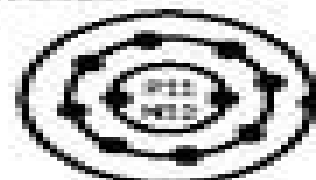


## Semester I Final Exam: Physical Science

Name: \_\_\_\_\_ Date: \_\_\_\_\_ Period: \_\_\_\_\_

Mark the correct answer or on your answer document.



1. What is the **atomic number** for the element above?

- a) 24
- b) 12
- c) 24.305
- d) Cannot be determined

2. What is the **atomic weight** for the element above?

- a) 24
- b) 12
- c) 24.305
- d) Cannot be determined

3. What is the **mass number** for the element above?

- a) 24
- b) 12
- c) 24.305
- d) Cannot be determined

4. What is the **charge** on an electron?

- a) negative
- b) positive
- c) neutral
- d) Cannot be determined

5. The number of **protons** in the nucleus of an atom determine the?

- a) Mass of the element
- b) The charge on the element
- c) The type of element
- d) Whether it is radioactive or not

6. Which **element** is being represented by the atom above?

- a) Mg (Magnesium)
- b) Na (Sodium)
- c) V (Vanadium)
- d) Na (Sodium)

7. What is the **outer electron** in the outer shell called?

- a) Single shell electron
- b) Valence electron
- c) Electron
- d) Neutral atom electron

8. What is the **mass number** for this atom?

- a) 20
- b) 22.9
- c) 11
- d) 12

9. The **atomic weight** is the average weight of all?

- a) Isotopes of that atom
- b) Protons in that atom
- c) Protons and electrons in that atom
- d) Neutrons in that atom

10. The periodic table of elements is **arranged in**.

- a) Atomic mass
- b) Atomic number
- c) Isotopes
- d) Numbers of electrons

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**Steven Giegerich**



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