Chapter 10 Review: Chemical Quantities & Calculations

(aka - mole & volume calculations, % composition, empirical & molecular formulas)

For the test, be sure that you are familiar, and able to complete, the problems below as well as the problems found on the following worksheets: "Molecular Weight and Mole Calculations," "Percent Comp. & Chemical Formula Calculations" and "Chapter 10 Sample Problems." Check the class website for answers to the review worksheet.

- The air around you is made up of 20.95% oxygen. How many oxygen molecules are in 75.0 grams of air? The density of air is 1.19 g/L. assume STP.
- How much helium (grams) would be needed to inflate a balloon to a volume of 5.50L? assume STP.
- The molecular formula of an antibacterial drug is C₁₇H₁₈FN₂O₃. How many fluorine atoms are in 150mg of this drug?

4. A molecule is made up of 40.0% carbon, 6.67% hydrogen, and an unknown amount of oxygen. What is the molecular formula of the molecule if its molar mass is 90.0g/mole?

- What is the empirical formula for a compound made up of 71.72% chlorine, 16.16% oxygen and 12.12% carbon? Calculate the molecular formula for the compound if it has a molar mass of 296,7g/mol.
- 6. How many molecules are in 8.42 Kilograms of C2H4?

Problems Review Chemical Quantities Practice

Rachel Sandford

Problems Review Chemical Quantities Practice:

Investigating Chemistry Matthew Johll, 2008-12-22 In its new second edition Investigating Chemistry A Forensic Science Perspective remains the only book that uses the inherently fascinating topics of crime and criminal investigations as a context for teaching the fundamental chemical concepts most often covered in an introductory nonmajors course Covering all the standard topics Matthew Johll capitalizes on the surge of interest in the scientific investigation of crime as sparked by CSI and other television shows bringing together the theme of forensic science and the fundamentals of chemistry in ways that are effective and accessible for students This edition features refined explanations of the chemical concepts which are the core of the book as well as a more thoroughly integrated forensic theme updated features and an expanded media NY Regents Chemistry Test Prep Review--Exambusters Flashcards Regents supplements package Exambusters, 2016-06-01 NY Regents CHEMISTRY Study Guide 700 questions and answers Essential definitions formulas concepts and sample problems Topics Introduction Matter Atoms Formulas Moles Reactions Elements Periodic Table Electrons Chemical Bonds Heat Gases Phase Changes Solutions Reaction Rates Equilibrium Acids and Bases Oxidation and Reduction Introduction to Organic Chemistry Radioactivity ADDITIONAL WORKBOOKS NY Regents INTEGRATED ALGEBRA Study Guide 450 questions and answers Essential definitions formulas concepts and sample problems Topics Sets Variables Exponents Properties of Numbers Like Terms Simple Equations Property of Equality Signed Numbers Monomials Polynomials Advanced Equations Verbal Problems Factoring Polynomials Algebraic Fractions Equations with Several Variables Advanced Verbal Problems Evaluating Formulas Simultaneous Equations Ratio and Proportion Variation Quadratic Equations and Radicals Coordinate Geometry

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