

## CHAPTER : 17

### CLASSIFICATION OF SOLIDS

(At Atomic level)

#### (i) Crystalline Solids :

- Solids in which atoms are arranged in pattern ; structure have arrangement.
- Arrangement of atoms repeats 3-dimensionally throughout the structure.

example : Metals : like copper  
Ionic compound : NaCl  
Ceramics : Zirconia.

Imp. MCQs

NOTE : Atoms arrangement in crystalline solids can be studied by x-rays diffraction using Bragg's equation

$$2d \sin \theta = m\lambda$$

Cohesive force between the atoms is v.v. large

Hardest

Melting Point  
v.v. high

- 
- sharp
  - definite

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