where k is the electrostatic force constant. The value of k depends on the nature of the medium between the two charges and the system of units chosen to measure  $F, q_1, q_2$  and r.

For two charges located in free space and in SI units, we have,

$$k = \frac{1}{4\pi\varepsilon_0} = 9x10^9 Nm^2 C^{-2} \tag{2}$$

where  $\varepsilon_0$  is the permittivity of free space.

#### Unit of charge:

1. In SI system: The SI unit of charge is coulomb. In the above equation, if  $q_1 = q_2 = 1C$  and r = 1m, then

$$F = \frac{1}{4\pi\varepsilon_0} = 9x10^9 N \tag{3}$$

Therefore, one coulomb is that amount of charge that repels an equal and similar charge with a force of  $9x10^9N$  when placed in vacuum at a distance of one metre from it.

- 2. In CGS system: The CGS unit of charge is stateoulomb. 1 coulomb = 3 x 10<sup>9</sup> stateoulomb = 3 x 10<sup>9</sup> e.s.u of charge. [e.s.u.= electrostatic unit] one e.s.u. or one stateoulomb is that charge which repels an identical charge in vacuum at a distance of 1 centimetre from it with a force of one dyne.
- Electromagnetic unit (e.m.u.): e.m.u. is the electromagnetic unit of charge or abcoulomb.

1 coulomb =  $\frac{1}{10}$  abcoulomb =  $\frac{1}{10}$  c.m.u. of charge.

Coulomb's law in vector form: In vector form coulomb's law may be expresses as,

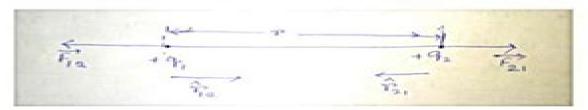


Figure 2: Two stationary point charge separated by a distance r (vector form).

 $\vec{F}_{21}$  = force on charge  $q_2$  due to charge  $q_1$ 

$$\vec{F}_{21} = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{r^2} r_{12}^2 \tag{4}$$

where,  $r_{12}^2 = \frac{r_{12}^2}{r}$ , is a unit vector in the direction from  $q_1$  to  $q_2$ . The coulombian force between unlike charges  $(q_1q_2 < 0)$  are attractive.

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