

QUESTION 1

1.1 Intermolecular forces are attractive forces between... (2)

- A. protons and electrons
- B. atoms
- C. molecules
- D. substances

1.2 What is the name of the polymer made of the monomer:
 $\text{CH}_2 = \text{CH}_2$ (2)

- A. Ethene
- B. Polyethene
- C. Polyester
- D. PVC

1.3 The following reaction occurs:



Which of the following can be used to measure the rate of this reaction? (2)

- A. Measuring the volume of gas produced per unit time
- B. Observing the precipitate formed
- C. Observing the changes in mass
- D. All of the above

1.4 The reaction below has reached equilibrium at a temperature of 50°C in a sealed gas syringe.



Consider the following statements:

- I. An increase in the amount of H_2S will favor the forward reaction and increase the K_c value
- II. An increase in pressure will not change the K_c -value and will not favor the forward or the reverse reaction.
- III. An increase in temperature will favor the forward reaction and

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Table of Contents Physical Science Paper Question Of Midyear Exam 24

1. Understanding the eBook Physical Science Paper Question Of Midyear Exam 24
 - The Rise of Digital Reading Physical Science Paper Question Of Midyear Exam 24
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Paper Question Of Midyear Exam 24
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Paper Question Of Midyear Exam 24
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Paper Question Of Midyear Exam 24

- Personalized Recommendations
 - Physical Science Paper Question Of Midyear Exam 24 User Reviews and Ratings
 - Physical Science Paper Question Of Midyear Exam 24 and Bestseller Lists
5. Accessing Physical Science Paper Question Of Midyear Exam 24 Free and Paid eBooks
 - Physical Science Paper Question Of Midyear Exam 24 Public Domain eBooks
 - Physical Science Paper Question Of Midyear Exam 24 eBook Subscription Services
 - Physical Science Paper Question Of Midyear Exam 24 Budget-Friendly Options
 6. Navigating Physical Science Paper Question Of Midyear Exam 24 eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science Paper Question Of Midyear Exam 24 Compatibility with Devices
 - Physical Science Paper Question Of Midyear Exam 24 Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Paper Question Of Midyear Exam 24
 - Highlighting and Note-Taking Physical Science Paper Question Of Midyear Exam 24
 - Interactive Elements Physical Science Paper Question Of Midyear Exam 24
 8. Staying Engaged with Physical Science Paper Question Of Midyear Exam 24
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Paper Question Of Midyear Exam 24
 9. Balancing eBooks and Physical Books Physical Science Paper Question Of Midyear Exam 24
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Paper Question Of Midyear Exam 24
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Physical Science Paper Question Of Midyear Exam 24
 - Setting Reading Goals Physical Science Paper Question Of Midyear Exam 24
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Physical Science Paper Question Of Midyear Exam 24

- Fact-Checking eBook Content of Physical Science Paper Question Of Midyear Exam 24
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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