PHYSICAL SCIENCES (CHEMISTRY) GRADE 11 TERM 1 FORMAL EXPERIMENT - PART 2 VERIFICATION OF BOYLE'S LAW MARKING GUIDELINES

1.1 • Aim stated correctly.
 e.g. To determine the relationship between the volume of a fixed

e.g. To determine the relationship between the volume of a fixed amount of gas and the pressure on the gas when temperature remains constant.

(1)

(5)

(65)

- 1.2 All apparatus listed ✓
 - 1. Boyle's Apparatus
 - 2. Pump
 - All steps of the method correct and logically stated, ✓✓ correct tense used. ✓✓

e.g.

- The tap on the Boyle's Law apparatus was opened and then the pump was connected.
- The apparatus was pumped to the pressure.
- 3. The oil was allowed to settle for a few minutes.
- A table was used to record the volume reading on the tube and the corresponding pressure reading on the Bourdon pressure gauge.
- The air tap was slightly opened and quickly closed to let a little air escape from the system. The oil was allowed to settle for a few minutes and then the pressure and volume readings were taken.
- 6. 5 different readings were recorded in the table.

2. Observations and Recording of results:

Learners will use results obtained during experiment to complete the table below.

Volume (V) cm ³	Pressure (p) kPa	pV (*)
1000		
	1	
		29

Marking guidelines:

- ✓ for correct values of the volume
- ✓ for correct values of the pressure
- ✓ ✓ for correct values of

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SJ Ball

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