



Problem Set 4
Chapters 4 and 25
Due Thursday Oct 24th

Nuclear Chemistry
Essay Due Oct 30th
Test Wednesday October 30th

- Your work is to be done on separate sheets of paper.
- All of the papers should be stapled together in the upper left hand corner with this question sheet at the back of the packet.
- Each of your responses should be numbered the same way as it appears on this packet.

Conceptual Questions

- 1) State the law of conservation of mass.
- 2) Why are atoms unstable?
- 3) How do unstable atoms gain stability?
- 4) A radioactive element having atomic number 82 and atomic mass 214 loses a beta particle, ${}^0_{-1}\text{e}$. What is the resulting particle?
- 5) What type of decay is the reaction below, and what isotope completes this nuclear equation?
 ${}^{27}_{13}\text{Al} + {}^4_2\text{He} \rightarrow {}^{31}_{15}\text{P} + ?$

Practice Questions

- 6) Complete this chart on the three types of radioactive decay

Type	Symbol	Mass (amu)	Charge
Alpha			
Beta			
Gamma			

- 7) If the radioactive atom ${}^{238}_{92}\text{U}$ emits an alpha particle, the atom remaining is represented by
 (A) ${}^{238}_{92}\text{U}$ (B) ${}^{236}_{92}\text{Th}$ (C) ${}^{234}_{92}\text{U}$ (D) ${}^{234}_{90}\text{Th}$ (E) ${}^{234}_{91}\text{Pa}$
- 8) What atom completes this equation? ${}^{30}_{14}\text{Si} + ? \rightarrow {}^{32}_{16}\text{S} + {}^4_2\text{He}$
 (A) ${}^4_2\text{He}$ (B) ${}^2_1\text{H}$ (C) ${}^1_0\text{n}$ (D) ${}^0_{-1}\text{e}$
- 9) What particle is emitted when lead-214, ${}^{214}_{82}\text{Pb}$, undergoes natural radioactive decay to bismuth-214, ${}^{214}_{83}\text{Bi}$?
 (A) an alpha particle, ${}^4_2\text{He}$ (C) a proton, ${}^1_1\text{H}$
 (B) a beta particle, ${}^0_{-1}\text{e}$ (D) a neutron, ${}^1_0\text{n}$

Extension Question

- 10) Recently Alexander Valterovich Litvinenko was poisoned by radioactive polonium-210.
 - a. Research who he was and why he was poisoned.
 - b. Find out what polonium-210 does to people.

Physical Scienceworkbook Chapter Nuclear Chemistry

Wordwise Definitions

Thomas Griffiths



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1. Understanding the eBook Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Definitions
 - The Rise of Digital Reading Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Definitions
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Definitions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Definitions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Definitions
 - Personalized Recommendations
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 - Highlighting and Note-Taking Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Definitions
 - Interactive Elements Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Definitions
- 8. Staying Engaged with Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Definitions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Definitions
- 9. Balancing eBooks and Physical Books Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Definitions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Definitions
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Definitions
 - Setting Reading Goals Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Definitions
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Definitions
 - Fact-Checking eBook Content of Physical Scienceworkbook Chapter Nuclear Chemistry Wordwise Definitions

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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