



basic education

Department:
Basic Education
REPUBLIC OF SOUTH AFRICA

PHYSICAL SCIENCES

GRADE 12

JUNE EXAMS- MEMO

2023

QUESTION 7 (start on a new page.)

- 7.1 Define an acid according to the Brønsted-Lowry theory.
- 7.2 The table below shows the ionisation constants (K_a) of two acids of equal concentrations.

NAME	FORMULA	K_a
Sulphuric acid	H_2SO_4	$1,3 \times 10^{-2}$
Sulphurous acid	H_2SO_3	$1,54 \times 10^{-2}$

- 7.2.1 Which ONE of these acids will have a higher molar conductivity? Give a reason for this answer.
- 7.2.2 Write down a chemical reaction to show the ionisation of sulphuric acid in water.

A	B
C CH_3Cl	D Propenoic acid
E Propanol	F $\text{C}_2\text{H}_5\text{Cl}$

2011

- 5.3.4 Explain the shape of the graph and what happened after t_0 . (2)

- 5.4 The following Maxwell-Boltzmann distribution graph was obtained at a low temperature.



- 5.4.1 Copy the graph given above in your ANSWER BOOK. Use a dotted line and indicate on the graph how the distribution would change at a higher TEMPERATURE.

20

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Ying-Ying Zheng

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Table of Contents Physical Sciences Paper June 23

1. Understanding the eBook Physical Sciences Paper June 23
 - The Rise of Digital Reading Physical Sciences Paper June 23
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Sciences Paper June 23
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Sciences Paper June 23
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Sciences Paper June 23
 - Personalized Recommendations
 - Physical Sciences Paper June 23 User Reviews and Ratings
 - Physical Sciences Paper June 23 and Bestseller Lists
5. Accessing Physical Sciences Paper June 23 Free and Paid eBooks
 - Physical Sciences Paper June 23 Public Domain eBooks
 - Physical Sciences Paper June 23 eBook Subscription Services
 - Physical Sciences Paper June 23 Budget-Friendly Options
6. Navigating Physical Sciences Paper June 23 eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Sciences Paper June 23 Compatibility with Devices
 - Physical Sciences Paper June 23 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Sciences Paper June 23
 - Highlighting and Note-Taking Physical Sciences Paper June 23
 - Interactive Elements Physical Sciences Paper June 23
8. Staying Engaged with Physical Sciences Paper June 23

- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Sciences Paper June 23
9. Balancing eBooks and Physical Books Physical Sciences Paper June 23
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Sciences Paper June 23
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Sciences Paper June 23
- Setting Reading Goals Physical Sciences Paper June 23
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Sciences Paper June 23
- Fact-Checking eBook Content of Physical Sciences Paper June 23
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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