# QUESTION 3 (Start on a new page)

3.4

Compounds A to E, indicated in the table below, are used during two investigations to determine the factors which influence boiling point.

Investigation	7	Compound	Molecular mass (g·mol·1)	Boiling point (°C)
	A	2,2-dimethyl propane	72	9
I i	В	2-methyl butane	72	27
	C	pentane	72	36
11	D	CH;CH;CH;CH;OH	74	117
	E	CH2CH2CH2CHO	72	75

3.1.1 Definition of the term structural isomers. Write down the:  3.1.2 GENERAL FORMULA of the homologous series to which these compounds belong  3.1.2 Type of structural isomerism illustrated by these compounds  3.2 Consider the boiling points of the compounds in investigation I.  3.2.1 Define the term boiling point.  3.2.2 Write down the independent variable for this investigation.  3.2.3 Write down one control variable for this investigation.  3.2.4 Explain fully why the boiling points increase from compound A to compound C.  3.2.5 Which one of compounds A or C will have the highest vapour pressure at a certain temperature?  Refer to the data in the table and give a reason for the answer.  3.3 To which homologous series does compound E belong?
3.1.2. GENERAL FORMULA of the homologous series to which these compounds belong 3.1.2. Type of structural isomerism illustrated by these compounds 3.2.1. Define the term boiling point. 3.2.1. Define the term boiling point. 3.2.2. Write down the independent variable for this investigation. 3.2.3. Write down one control variable for this investigation. 3.2.4. Explain fully why the boiling points increase from compound A to compound C. 3.2.5. Which one of compounds A or C will have the highest vapour pressure at a certain temperature? Refer to the data in the table and give a reason for the answer.
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pressure at a certain temperature? Refer to the data in the table and give a reason for the answer.
3.3 To which homologous series does compound E belong?

Consider investigation II. Refer to the type of Van Der Waals forces in each of the compounds and explain why the boiling point of compound D is

higher than that of compound E.

(3)

(3) [17]

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# **Table of Contents Physical Science Question Papers Grade 12 2011**

- 1. Understanding the eBook Physical Science Question Papers Grade 12 2011
  - The Rise of Digital Reading Physical Science Question Papers Grade 12 2011
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Physical Science Question Papers Grade 12 2011
  - Exploring Different Genres
  - o Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Science Question Papers Grade 12 2011
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Physical Science Question Papers Grade 12 2011
  - Personalized Recommendations
  - Physical Science Question Papers Grade 12 2011 User Reviews and Ratings

- Physical Science Question Papers Grade 12 2011 and Bestseller Lists
- 5. Accessing Physical Science Question Papers Grade 12 2011 Free and Paid eBooks
  - Physical Science Question Papers Grade 12 2011 Public Domain eBooks
  - Physical Science Question Papers Grade 12 2011 eBook Subscription Services
  - o Physical Science Question Papers Grade 12 2011 Budget-Friendly Options
- 6. Navigating Physical Science Question Papers Grade 12 2011 eBook Formats
  - o ePub, PDF, MOBI, and More
  - Physical Science Question Papers Grade 12 2011 Compatibility with Devices
  - Physical Science Question Papers Grade 12 2011 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Physical Science Question Papers Grade 12 2011
  - Highlighting and Note-Taking Physical Science Question Papers Grade 12 2011
  - o Interactive Elements Physical Science Question Papers Grade 12 2011
- 8. Staying Engaged with Physical Science Question Papers Grade 12 2011
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physical Science Question Papers Grade 12 2011
- 9. Balancing eBooks and Physical Books Physical Science Question Papers Grade 12 2011
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical Science Question Papers Grade 12 2011
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Question Papers Grade 12 2011
  - Setting Reading Goals Physical Science Question Papers Grade 12 2011
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Question Papers Grade 12 2011
  - Fact-Checking eBook Content of Physical Science Question Papers Grade 12 2011
  - Distinguishing Credible Sources

- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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