

- 3.6 What type of chemical bond is present in NH_3 ? Draw Lewis electron dot diagram of NH_3 .
(The atomic number of H and N are 1 and 7 respectively) 2
- 3.7 How is ionic bond formed in sodium fluoride?
(The atomic number of F and Na are 9 and 11 respectively) 2
- OR**
- C has 4 electrons in its outermost shell and O has 6 electrons in its outermost shell. Draw the Lewis dot diagram of CO_2 molecule. 2
- 3.8 Distinguish between naphthalene and sodium chloride by two physical properties <https://www.westbengalboard.com> 2
- 3.9 What is meant by strong electrolyte? 2
- OR**
- What are meant by cathode and anode electrodes? 2

Group 'D'

4. Answer the following questions (alternatives are to be noted) :
- 4.1 Establish the combined form of Boyle's law and Charles' law. 3
- 4.2 By the reaction of aluminium with 40 grams of metallic oxide and high temperature, 28 grams of the metal and 25.5 grams of aluminium oxide are produced. How many grams of aluminium are required for the reaction? 3
- OR**
- The chemical equation for the reaction between sulphuric acid and sodium hydroxide is:
- $$\text{H}_2\text{SO}_4 + 2\text{NaOH} \longrightarrow \text{Na}_2\text{SO}_4 + 2\text{H}_2\text{O}$$
- How many grams of sodium hydroxide will be required to react completely with 4.9 grams of sulphuric acid?
(H = 1, O = 16, Na = 23, S = 32) 3
- 4.3 A light is incident from air medium on another medium. If the refractive index of this medium is 1.5 and the wavelength of light in this medium is 4000\AA , what is the wavelength of light in the air medium? What is the velocity of the light in this medium? 3

OR

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M Tight



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Table of Contents Physical Science Question Paper 2mid Year

1. Understanding the eBook Physical Science Question Paper 2mid Year
 - The Rise of Digital Reading Physical Science Question Paper 2mid Year
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Question Paper 2mid Year
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Question Paper 2mid Year
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Question Paper 2mid Year
 - Personalized Recommendations
 - Physical Science Question Paper 2mid Year User Reviews and Ratings
 - Physical Science Question Paper 2mid Year and Bestseller Lists
5. Accessing Physical Science Question Paper 2mid Year Free and Paid eBooks
 - Physical Science Question Paper 2mid Year Public Domain eBooks

- Physical Science Question Paper 2mid Year eBook Subscription Services
- Physical Science Question Paper 2mid Year Budget-Friendly Options
- 6. Navigating Physical Science Question Paper 2mid Year eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science Question Paper 2mid Year Compatibility with Devices
 - Physical Science Question Paper 2mid Year Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Question Paper 2mid Year
 - Highlighting and Note-Taking Physical Science Question Paper 2mid Year
 - Interactive Elements Physical Science Question Paper 2mid Year
- 8. Staying Engaged with Physical Science Question Paper 2mid Year
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Question Paper 2mid Year
- 9. Balancing eBooks and Physical Books Physical Science Question Paper 2mid Year
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Question Paper 2mid Year
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Question Paper 2mid Year
 - Setting Reading Goals Physical Science Question Paper 2mid Year
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Question Paper 2mid Year
 - Fact-Checking eBook Content of Physical Science Question Paper 2mid Year
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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