

L. N. Coakley Science Department
Specific Heat Capacity Worksheet.

Name: _____ Date: _____ Grade: _____.

Key points to remember:

The specific heat capacity of a substance is the heat required to produce a 1 °C rise in 1 kg.

The heat received or given out = mass x specific heat capacity x temp. change

$$Q = mc\Delta T$$

$$c = \frac{Q}{m\Delta T}$$

1. How much heat is need it to raise the temperature by 10 °C of 5 kg of a substance of specific heat capacity 300 J/(kg °C)?

Known:

Formula:

Substitution:

$\Delta T =$

$m =$

$c =$

Unknown:

The amount of heat need it is _____ J.

2. The same amount of heat was given to different masses of three substances A, B and C. The temperature rise in each case is shown in the table. Calculate the specific heat capacity of A, B, and C.

Material	Mass (kg)	Heat given (J)	Temp. rise (°C)
A	1.0	2000	1.0
B	2.0	2000	5.0
C	0.5	2000	4.0

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Table of Contents Physical Science Pearson Using Heat Answers

1. Understanding the eBook Physical Science Pearson Using Heat Answers
 - The Rise of Digital Reading Physical Science Pearson Using Heat Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Pearson Using Heat Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Pearson Using Heat Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Pearson Using Heat Answers
 - Personalized Recommendations
 - Physical Science Pearson Using Heat Answers User Reviews and Ratings
 - Physical Science Pearson Using Heat Answers and Bestseller Lists
5. Accessing Physical Science Pearson Using Heat Answers Free and Paid eBooks

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 - Physical Science Pearson Using Heat Answers eBook Subscription Services
 - Physical Science Pearson Using Heat Answers Budget-Friendly Options
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 - Physical Science Pearson Using Heat Answers Enhanced eBook Features
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 - Highlighting and Note-Taking Physical Science Pearson Using Heat Answers
 - Interactive Elements Physical Science Pearson Using Heat Answers
8. Staying Engaged with Physical Science Pearson Using Heat Answers
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Pearson Using Heat Answers
9. Balancing eBooks and Physical Books Physical Science Pearson Using Heat Answers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Pearson Using Heat Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Pearson Using Heat Answers
- Setting Reading Goals Physical Science Pearson Using Heat Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Pearson Using Heat Answers
- Fact-Checking eBook Content of Physical Science Pearson Using Heat Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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