

Question 1: (8 questions) (multiple choice)

Q1. Which of the following statements are correct? Select all that apply. ☐ 1 ☒ 2 ☐ 3 ☐ 4

Q2. Which of the following is not a valid method of determining the concentration of a solution? ☐ titration

Q3. Which of the following are the types of ligands in the complex [Co(NH₃)₄(H₂O)₂Cl]³⁺? Select all that apply. ☐ monodentate ☐ bidentate ☐ tridentate ☐ tetradentate

Number

Answer 1: 1, 2

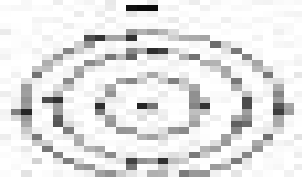
Answer 2: 1, 2, 3

Answer 3: 1, 2, 3, 4

Answer 4: 1, 2, 3

Answer 5: 1, 2, 3

Number of ligands



Number of ligands



Q4. Consider the following octahedral complexes. Which of the following are not true? Select all that apply.



Li

Number of ligands

Answer 1: 1

Answer 2: 1

Answer 3: 1

Answer 4: 1



Na

Number of ligands

Answer 1: 1

Answer 2: 1

Answer 3: 1

Answer 4: 1



Mg

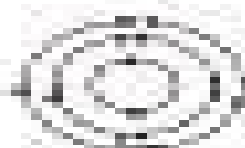
Number of ligands

Answer 1: 1

Answer 2: 1

Answer 3: 1

Answer 4: 1



Cl

Number of ligands

Answer 1: 1

Answer 2: 1

Answer 3: 1

Answer 4: 1



He

Number of ligands

Answer 1: 1

Answer 2: 1

Answer 3: 1

Answer 4: 1



Se

Number of ligands

Answer 1: 1

Answer 2: 1

Answer 3: 1

Answer 4: 1

Q5. Which of the following statements are correct? Select all that apply.

Q6. Which of the following is a valid method of determining the concentration of a solution? ☐ titration

Q7. Which of the following are the types of ligands in the complex [Co(NH₃)₄(H₂O)₂Cl]³⁺? ☐ monodentate ☐ bidentate ☐ tridentate ☐ tetradentate

Q8. Which of the following are the types of ligands in the complex [Co(NH₃)₄(H₂O)₂Cl]³⁺? ☐ monodentate ☐ bidentate ☐ tridentate ☐ tetradentate

Q9. Which of the following are the types of ligands in the complex [Co(NH₃)₄(H₂O)₂Cl]³⁺? ☐ monodentate ☐ bidentate ☐ tridentate ☐ tetradentate

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