



CERTIFICATE NOT ISSUED

SM RA2021290000282

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 ISSUED: _____ EYES: _____
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 FEE: 10.00 (C) ADD GENDER: _____

THIS PERMIT ONLY VALID WHEN USED IN COMPLIANCE
 WITH THE PROVISIONS SET FORTH ON NJSA 39:3-13.2
 NJSA 39:3-13.2 AND NJSA 39:3-13.2A
 REVISED STATUTES

Road Test
Location, Date, & Time

REJECTED	BY	RETURN
_____	_____	_____
_____	_____	_____

APPT FOR DRIVING TEST
 DATE _____ TIME _____
 ACTUAL _____

RESTRICTIONS _____
 DRIVING _____ DATE _____ X _____

KNOWLEDGE TEST:

AUTO _____

THIS STUDENT HAS SUCCESSFULLY
 COMPLETED THE BEHIND THE WHEEL
 COURSE.

Road Rules
Driving School

NAME OF SCHOOL _____

PRINCIPAL OR OWNER OF
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Nj Motorcycle License Written Test

Thomas Griffiths



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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3.655
furthermore using the stoichiometric ratio between the naoh and the acid analyte 1:1 we could experimentally determine the number of moles of the acid in the solution to be 0.00731 moles making the experimental molar mass

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 $HCl(aq) + NaOH(aq) \rightarrow NaCl(aq) + H_2O(l)$
one mole of HCl would be fully neutralized by one mole of NaOH

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a pH titration is used to determine the concentration of an acid or a base titrations play an important role in determining amount and purity in many manufacturing processes

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