

Section 7.4 - Determining Chemical Formulas

Calculations of Molecular Formulas

An empirical formula may or may not be a correct molecular formula.

Book example , diborane¹'s empirical formula is BH_3 , any multiple of that equals the same ratio - $\text{B}_2\text{H}_6, \text{B}_3\text{H}_9, \text{B}_4\text{H}_{12}$ etc

It is a colorless gas at room temperature with a repulsively sweet odor.
Diborane mixes well with air, easily forming explosive mixtures.
Diborane will ignite spontaneously in moist air at room temperature.

Section Quiz Determining Chemical Formulas

H. Scott Fogler



Section Quiz Determining Chemical Formulas:

Chemistry All-in-One For Dummies (+ Chapter Quizzes Online) Christopher R. Hren, John T. Moore, Peter J. Mikulecky, 2022-11-23 Everything you need to crush chemistry with confidence Chemistry All in One For Dummies arms you with all the no nonsense how to content you ll need to pass your chemistry class with flying colors You ll find tons of practical examples and practice problems and you ll get access to an online quiz for every chapter Reinforce the concepts you learn in the classroom and beef up your understanding of all the chemistry topics covered in the standard curriculum Prepping for the AP Chemistry exam Dummies has your back with plenty of review before test day With clear definitions concise explanations and plenty of helpful information on everything from matter and molecules to moles and measurements Chemistry All in One For Dummies is a one stop resource for chem students of all valences Review all the topics covered in a full year high school chemistry course or one semester of college chemistry Understand atoms molecules and the periodic table of elements Master chemical equations solutions and states of matter Complete practice problems and end of chapter quizzes online Chemistry All In One For Dummies is perfect for students who need help with coursework or want to cram extra hard to ace that chem test

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Creating Wicked Students Paul Hanstedt, 2023-07-03 In Creating Wicked Students Paul Hanstedt argues that courses can and should be designed to present students with what are known as wicked problems because the skills of dealing with such knotty problems are what will best prepare them for life after college. As the author puts it this book begins with the assumption that what we all want for our students is that they be capable of changing the world. When a student leaves college we want them to enter the world not as drones participating mindlessly in activities to which they've been appointed but as thinking deliberative beings who add something to society. There's a lot of talk in education these days about wicked problems problems that defy traditional expectations or knowledge problems that evolve over time. Zika ISIS political discourse in the era of social media. To prepare students for such wicked problems they need to have wicked

competencies the ability to respond easily and on the fly to complex challenges Unfortunately a traditional education that focuses on content and skills often fails to achieve this sense of wickedness Students memorize for the test prepare for the paper practice the various algorithms over and over again but when the parameters or dynamics of the test or the paper or the equation change students are often at a loss for how to adjust This is a course design book centered on the idea that the goal in the college classroom in all classrooms all the time is to develop students who are not just loaded with content but capable of using that content in thoughtful deliberate ways to make the world a better place Achieving this goal requires a top to bottom reconsideration of courses including student learning goals text selection and course structure day to day pedagogies and assignment and project design Creating Wicked Students takes readers through each step of the process providing multiple examples at each stage while always encouraging instructors to consider concepts and exercises in light of their own courses and students Bulletin Carnegie Foundation for the Advancement of Teaching,1917

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