

Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers? *Frank can share his red marker with Flora and Flora can share her blue marker with Frank.*
2. Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set? *Olivia can share her purple and green markers with Otto and Otto can share his black and brown markers with Olivia.*

Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the *Covalent Bond's Gizmo*, atoms form bonds in this way.



To begin, check that **Fluorine** is selected from the **Select a substance** menu. Click **Play** () to see the electrons orbiting the nucleus of each atom.

1. The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have? *Each fluorine atom has seven valence electrons.*
2. Click **Pause** (). Drag an electron from the left atom to the right atom. Click **Play**. What happens? *The electrons that was dragged now orbits both atoms.*
3. Click **Pause**, drag an electron from the right atom to the left, and then click **Play**. What happens now? *A pair of electrons now orbit both atoms.*

Section 8 1 Forming Chemical Bonds Answer Key

Mike Jess



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Table of Contents Section 8 1 Forming Chemical Bonds Answer Key

1. Understanding the eBook Section 8 1 Forming Chemical Bonds Answer Key
 - The Rise of Digital Reading Section 8 1 Forming Chemical Bonds Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Section 8 1 Forming Chemical Bonds Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Section 8 1 Forming Chemical Bonds Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Section 8 1 Forming Chemical Bonds Answer Key
 - Personalized Recommendations
 - Section 8 1 Forming Chemical Bonds Answer Key User Reviews and Ratings
 - Section 8 1 Forming Chemical Bonds Answer Key and Bestseller Lists
5. Accessing Section 8 1 Forming Chemical Bonds Answer Key Free and Paid eBooks
 - Section 8 1 Forming Chemical Bonds Answer Key Public Domain eBooks
 - Section 8 1 Forming Chemical Bonds Answer Key eBook Subscription Services
 - Section 8 1 Forming Chemical Bonds Answer Key Budget-Friendly Options
6. Navigating Section 8 1 Forming Chemical Bonds Answer Key eBook Formats

- ePub, PDF, MOBI, and More
 - Section 8 1 Forming Chemical Bonds Answer Key Compatibility with Devices
 - Section 8 1 Forming Chemical Bonds Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Section 8 1 Forming Chemical Bonds Answer Key
 - Highlighting and Note-Taking Section 8 1 Forming Chemical Bonds Answer Key
 - Interactive Elements Section 8 1 Forming Chemical Bonds Answer Key
8. Staying Engaged with Section 8 1 Forming Chemical Bonds Answer Key
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Section 8 1 Forming Chemical Bonds Answer Key
9. Balancing eBooks and Physical Books Section 8 1 Forming Chemical Bonds Answer Key
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Section 8 1 Forming Chemical Bonds Answer Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Section 8 1 Forming Chemical Bonds Answer Key
- Setting Reading Goals Section 8 1 Forming Chemical Bonds Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Section 8 1 Forming Chemical Bonds Answer Key
- Fact-Checking eBook Content of Section 8 1 Forming Chemical Bonds Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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