

Periodic Table Named-Groups Worksheet – 2

DIRECTIONS: Use the Periodic Table shown below to answer each question. using the letters in each "box" to identify the required element(s).

The diagram shows a 10x10 grid representing a 2D plane. The grid is divided into four quadrants by a horizontal line and a vertical line. The top-left quadrant is labeled 'A' in the top-left corner. The top-right quadrant is labeled 'B' in the top-right corner. The bottom-left quadrant is labeled 'C' in the bottom-left corner. The bottom-right quadrant is labeled 'D' in the bottom-right corner. The grid is further divided into smaller sections by additional lines. Arrows point from the labels 'A', 'B', 'C', and 'D' to their respective sections.

List elements in *alphabetical order* when more than 1 element answers a question.

- _____ 1) Element **A** is in a box all by itself because: **A)** it is really a Noble Gas **B)** it is a gas **C)** it is NOT a gas **D)** it does NOT share the properties of the other elements in its group **E)** it is a liquid
- _____ 2) Which of these elements is/are highly reactive? **A)** "C" **B)** "E" **C)** "K" **D)** A, B, and C **E)** A and B only
- _____ 3) What is the letter of the element whose atomic number is closest to the element labeled "Z"?
- _____ 4) Which letter labels the halogen element with the lowest atomic number?
- _____ 5) Which of these elements is the least reactive? **A)** "J" **B)** "Y" **C)** "A" **D)** "O" **E)** "B"
- _____ 6) Which of these elements is/are NOT transition elements/metals. (list ALL that are not transition elements)? **A)** "C" **B)** "N" **C)** "K" **D)** "N" **E)** "Y"

continued on other side

Section 5 2 The Modern Periodic Table Worksheet Answers

SA Adler



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