

Density Worksheet
Physical Science
 $D = m/V$

Densities of Common Substances @ 20°C				
Substance	Density (g/cm ³)		Substance	Density (g/cm ³)
Oxygen	0.00133		Aluminum	2.70
Hydrogen	0.000084		Iron	7.87
Ethanol	0.785		Copper	8.96
Benzene	0.880		Silver	10.5
Water	1.000		Lead	11.34
Magnesium	1.74		Mercury	13.6
Salt (sodium chloride)	2.16		Gold	19.32

1. The ratio of an object's mass to its _____ is called the *density* of the object.
2. A kilogram of lead occupies a much smaller volume than a kilogram of water, because _____ has a much higher *density*.
3. For the masses and volumes indicated, calculate the **density** in grams per cubic centimeters.
 - A. mass = 453 g; volume = 225 cm³
 - B. mass = 5.0 g; volume = 10.0 cm³
 - C. mass = 26.1 g; volume = 2.0 mL
4. If 89.2 mL of a liquid has a mass of 75.2 g, calculate the liquid's density.
5. A cube of metal weighs 1450 g and displaces 542 mL of water when immersed. Calculate the density of the metal.

Section 8 1 Formation And Solutions Answers Physical Science

G Orfield



Section 8.1 Formation And Solutions Answers Physical Science:

Section8 1 Formation And Solutions Answers Physical Science Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Section8 1 Formation And Solutions Answers Physical Science**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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