

Stoichiometry

Quiz 11*9 Section 9

Section 9: Introduction to Stoichiometry

In the space provided, write the letter of the best response that best completes each problem or best answers each question.

1. Stoichiometry is the study of chemistry that deals with, depending on context, one or more chemical relationships.
 - a. balancing
 - b. identifying
 - c. energy conversion
 - d. mass relationships
 - e. physical measurements
2. Some basic units for energy are listed in a table of fundamental units.
 - a. the base unit measurement of energy
 - b. the total gas law
 - c. the base unit measurement of measurement
 - d. temperature-related
3. The number of significant figures in a product or quotient is determined by the number of significant figures in the least number of significant figures in the factors.
 - a. the number of significant figures in the least number of significant figures in the factors
 - b. the number of significant figures in the product or quotient
 - c. the number of significant figures in the product or quotient
 - d. the number of significant figures in the product or quotient
4. The chemical equation $\text{C}_2\text{H}_6 + 3\text{O}_2 \rightarrow 2\text{CO}_2 + 3\text{H}_2\text{O}$ is balanced.
 - a. 1 mole of C_2H_6 reacts with 3 moles of O_2 to form 2 moles of CO_2
 - b. 1 mole of C_2H_6 reacts with 3 moles of O_2 to form 2 moles of CO_2
 - c. 1 mole of C_2H_6 reacts with 3 moles of O_2 to form 2 moles of CO_2
 - d. 1 mole of C_2H_6 reacts with 3 moles of O_2 to form 2 moles of CO_2
5. The number of significant figures in the product of two numbers is determined by the number of significant figures in the least number of significant figures in the factors.
 - a. 1
 - b. 2
 - c. 3
 - d. 4

Section Quiz Introduction To Stoichiometry

RJ Shavelson



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