



Chemical Bonding



Name _____ Period: _____ Date: _____

CHEMICAL BONDS & VALENCE ELECTRONS

- 1) _____ = an interaction between atoms that holds them together by reducing the potential energy of their electrons
- 2) _____ = a shorthand representation of the composition of a substance using atomic symbols and numeric subscripts (Ex: NaCl, MgCl_2 , H_2O , $\text{C}_6\text{H}_{12}\text{O}_6$)
- 3) Chemical bonding involves the valence electrons of the atoms involved in a _____.
- 4) _____ = the electrons available to be lost, gained, or shared in the formation of chemical bonds
 - For atoms of main group elements, valence electrons are electrons in the _____ main energy level.
 - Use of the periodic table to determine the number of valence electrons for atoms of the main group elements:

Group	Number of Valence Electrons
1	
2	
13	
14	
15	
16	
17	
18	

_____ = a helium atom has _____ valence electrons

Section 8 Covalent Bonds Answer Key

The Princeton Review



Section 8 Covalent Bonds Answer Key:

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