

Acids and Bases Worksheet

1. Acids and bases share some properties but are different in others. Use the samples at the side of the class to complete the chart. If both acids and bases share the property, place it in both columns.

Property	Acids	Bases
Conductivity in Aqueous Solution	very	very
Taste	Sour	Bitter
Feel	—	Slippery
Reaction with Metal	produces H_2 gas	—
Reaction with Carbonate	produces CO_2 gas	—
Colour on Litmus Paper	Red	Blue
Colour with Universal Indicator	Red	Purple
Colour with phenolphthalein	Colourless	pink

2. Indicate if each compound is an acid or base and name it.

	Acid/Base?	Name
HNO_3	<u>A</u>	<u>nitric acid</u>
$Mg(OH)_2$	<u>B</u>	<u>magnesium hydroxide</u>
H_3PO_4	<u>A</u>	<u>phosphoric acid</u>
H_2CO_3	<u>A</u>	<u>carbonic acid</u>
HF	<u>A</u>	<u>hydrofluoric acid</u>
$Fe(OH)_3$	<u>B</u>	<u>iron (iii) hydroxide</u>

3. Write the formula for each of the following acids/bases:

hydrochloric acid	<u>HCl</u>	hydrochloric acid	<u>HCl</u>
aluminum hydroxide	<u>$Al(OH)_3$</u>	sodium hydroxide	<u>NaOH</u>
lithium hydroxide	<u>LiOH</u>	calcium hydroxide	<u>$Ca(OH)_2$</u>
sulfuric acid	<u>H_2SO_4</u>	sulfurous acid	<u>H_2SO_3</u>

Section 8 3 Properties Of Acids And Bases Worksheet Answers

Raffaela Di Napoli



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Table of Contents Section 8 3 Properties Of Acids And Bases Worksheet Answers

1. Understanding the eBook Section 8 3 Properties Of Acids And Bases Worksheet Answers
 - The Rise of Digital Reading Section 8 3 Properties Of Acids And Bases Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Section 8 3 Properties Of Acids And Bases Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Section 8 3 Properties Of Acids And Bases Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Section 8 3 Properties Of Acids And Bases Worksheet Answers
 - Personalized Recommendations
 - Section 8 3 Properties Of Acids And Bases Worksheet Answers User Reviews and Ratings

- Section 8 3 Properties Of Acids And Bases Worksheet Answers and Bestseller Lists
- 5. Accessing Section 8 3 Properties Of Acids And Bases Worksheet Answers Free and Paid eBooks
 - Section 8 3 Properties Of Acids And Bases Worksheet Answers Public Domain eBooks
 - Section 8 3 Properties Of Acids And Bases Worksheet Answers eBook Subscription Services
 - Section 8 3 Properties Of Acids And Bases Worksheet Answers Budget-Friendly Options
- 6. Navigating Section 8 3 Properties Of Acids And Bases Worksheet Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Section 8 3 Properties Of Acids And Bases Worksheet Answers Compatibility with Devices
 - Section 8 3 Properties Of Acids And Bases Worksheet Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Section 8 3 Properties Of Acids And Bases Worksheet Answers
 - Highlighting and Note-Taking Section 8 3 Properties Of Acids And Bases Worksheet Answers
 - Interactive Elements Section 8 3 Properties Of Acids And Bases Worksheet Answers
- 8. Staying Engaged with Section 8 3 Properties Of Acids And Bases Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Section 8 3 Properties Of Acids And Bases Worksheet Answers
- 9. Balancing eBooks and Physical Books Section 8 3 Properties Of Acids And Bases Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Section 8 3 Properties Of Acids And Bases Worksheet Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Section 8 3 Properties Of Acids And Bases Worksheet Answers
 - Setting Reading Goals Section 8 3 Properties Of Acids And Bases Worksheet Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Section 8 3 Properties Of Acids And Bases Worksheet Answers
 - Fact-Checking eBook Content of Section 8 3 Properties Of Acids And Bases Worksheet Answers
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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