# Chapter 4

# Section 2 A Guided Tour of the Periodic Table

# **Objectives**

- Relate the organization of the periodic table to the arrangement of electrons within an atom.
- Explain why some atoms gain or lose electrons to form ions.
- Determine how many protons, neutrons, and electrons an atom has, given its symbol, atomic number, and mass number.
- Describe how the abundance of isotopes affects an element's average atomic mass.

Chapter menu

Resources

# **Section Tour Of The Periodic Table**

**CO Houle** 

#### **Section Tour Of The Periodic Table:**

Inorganic Chemistry Mark Weller, Mark T. Weller, Tina Overton, Jonathan Rourke, Fraser Armstrong, 2014 Leading the reader from the fundamental principles of inorganic chemistry right through to cutting edge research at the forefront of the subject Inorganic Chemistry Sixth Edition is the ideal course companion for the duration of a student's degree The authors have drawn upon their extensive teaching and research experience in updating this established text the sixth edition retains the much praised clarity of style and layout from previous editions while offering an enhanced Frontiers section Exciting new applications of inorganic chemistry have been added to this section in particular relating to materials chemistry and medicine This edition also sees a greater use of learning features to provide students with all the support they need for their studies Providing comprehensive coverage of inorganic chemistry while placing it in context this text will enable the reader to fully master this important subject Online Resource Centre For registered adopters of the text Figures marginal structures and tables of data ready to download Test bank For students Answers to self tests and exercises from the book Videos of chemical reactions Tables for group theory Web links Interactive structures and other resources on www chemtube3D com **Elements** Adrian Dingle, 2017 Science Tutor: Chemistry, Grades 7 - 8 Raham, 2008-08-28 Connect students in grades 7 and up with science using Science Tutor Chemistry This effective 48 page resource provides additional concept reinforcement for students who struggle in chemistry Each lesson in this book contains an Absorb section to instruct and simplify concepts and an Apply section to help students grasp concepts on their own The book covers topics such as matter physical and chemical changes mixtures and solutions the periodic table atomic structure and radioactivity It is great for use in the classroom and at home Science Tutor: Chemistry, Grades 7 - 12 Gary Raham, 2005-01-03 Introduces new chemistry concepts and provides activities so that students can practice and grasp the concepts Key terms are highlighted in the text as well as in a comprehensive glossary Answer keys are included Shriver and Atkins' Inorganic Chemistry Peter Atkins, 2010 Inorganic Chemistry fifth edition represents an integral part of a student's chemistry education Basic chemical principles are set out clearly in Foundations and are fully developed throughout the text culminating in the cutting edge research topics of the Frontiers which illustrate the dynamic nature of inorganic chemistry Chemistry for Nerds Guide Book: Chemistry, Science, Nerd, Geek, Textbook, Guidebook, Study Guide, Educational, STEM, Science Gift Matt Kingsley, Calling all curious minds and science enthusiasts Are you fascinated by the invisible forces that shape our world Do you crave a deeper understanding of the elements molecules and reactions that make up everything around us Then look no further than Chemistry for Nerds Unleash Your Inner Mad Scientist This isn t your typical boring textbook This is a thrilling adventure through the captivating world of chemistry written in a way that s engaging accessible and downright fun Inside these pages you ll discover The secrets of matter From atoms and molecules to the states of matter and the laws that govern them The magic of reactions Explore the explosive world of chemical reactions from kinetics and equilibrium to acids bases

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