

11.1

DESCRIBING CHEMICAL REACTIONS

Section Review

Objectives

- 1. Represent the processes of a reaction.
- 2. Explain how to write a balanced equation.
- 3. List the steps for writing a complete ionic equation.

Vocabulary

- 1. chemical equation
- 2. balanced equation
- 3. reactant
- 4. coefficient
- 5. product

Part A Completion

Write the coefficient for each in the balanced equation, and balance the equation by changing the coefficient for each substance. Write balanced after the equation. (Write the coefficient 1.)

1. A chemical reaction is usually represented by a balanced _____ 1. _____
 The chemical formula for a chemical compound is _____ 2. _____ The mass
 number for an element is represented by the _____ 3. _____ An element's atomic
 mass is represented by _____ 4. _____ A chemical reaction between two or
 more substances is called a _____ 5. _____ or chemical reaction. The
 products are the substances that are formed. The _____ 6. _____ of a reaction is the
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1. _____
2. _____
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6. _____
7. _____
8. _____
9. _____
10. _____

Section Review 11 1 Describing Chemical Reactions

Brendan G. Carr



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