

Element	symbol	Atomic number	Electronic arrangement in different orbit					
			K	L	M	N	O	P
Hydrogen	H	1	1					
Helium	He	2	2					
Lithium	Li	3	2	1				
Beryllium	Be	4	2	2				
Boron	B	5	2	3				
Carbon	C	6	2	4				
Nitrogen	N	7	2	5				
Oxygen	O	8	2	6				
Fluorine	F	9	2	7				
Neon	Ne	10	2	8				
Sodium	Na	11	2	8	1			
Magnesium	Mg	12	2	8	2			
Aluminium	Al	13	2	8	3			
Silicon	Si	14	2	8	4			
Phosphorus	P	15	2	8	5			
Sulphur	S	16	2	8	6			
Chlorine	Cl	17	2	8	7			
Argon	Ar	18	2	8	8			
Potassium	K	19	2	8	8	1		
Calcium	Ca	20	2	8	8	2		

● $c = v\lambda$

● $\bar{v} = \frac{1}{\lambda} = \frac{v}{c}$

[c = velocity of light in free space,

v = frequency, λ = wavelength]

[$\bar{v}$ = wave number]

Physical Science Notes Chapter Atomic Structure

Bishopsgate Institute, London



Physical Science Notes Chapter Atomic Structure:

Physical Science Notes Chapter Atomic Structure Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Physical Science Notes Chapter Atomic Structure**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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