

Physical Science
Chapter 2

Properties of Matter

Classifying Matter

Substances

- **Element:** a substance that can't be separated down into simpler substances.
- **Atom:** the smallest particle of an element.
- An element has a fixed composition because it contains only one type of atom.

2 Types of Mixtures

1. **Heterogeneous Mixture:** have materials that are not evenly distributed and noticeably different from one another.
 2. **Homogeneous Mixture:** have materials that are evenly distributed so it is difficult to distinguish one substance from another.
 - a. Also called a solution
 - b. Remain constantly and uniformly mixed
- Which type of mixture is milk?
 - Based on the size of its largest particles, a mixture can be classified as a solution, a suspension, or a colloid.

Solutions

Solution: when substances dissolve and form a homogenous mixture.

- Examples: windshield wiper fluid, sea water, lemonade

Colloids

Colloid: contains some particles that are larger than those in a solution, but smaller than those in a suspension.

- Examples: homogenized milk, fog
- Think about it...
 - What happens to the size of the drops of fat in milk when milk is homogenized?

The Scattering of Light

- Colloids and Suspensions scatter light just like fog.
- Light goes right through a solution.

Chemical Properties

- Any ability to produce a change in the composition of matter
- Chemical properties can be observed only when the substance in a sample of matter is changing into different substances.

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1. Understanding the eBook Physical Science Chapter 2 Review Answers Properties Of Matter
 - The Rise of Digital Reading Physical Science Chapter 2 Review Answers Properties Of Matter
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Chapter 2 Review Answers Properties Of Matter
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Chapter 2 Review Answers Properties Of Matter
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Chapter 2 Review Answers Properties Of Matter
 - Personalized Recommendations
 - Physical Science Chapter 2 Review Answers Properties Of Matter User Reviews and Ratings
 - Physical Science Chapter 2 Review Answers Properties Of Matter and Bestseller Lists
5. Accessing Physical Science Chapter 2 Review Answers Properties Of Matter Free and Paid eBooks
 - Physical Science Chapter 2 Review Answers Properties Of Matter Public Domain eBooks
 - Physical Science Chapter 2 Review Answers Properties Of Matter eBook Subscription Services
 - Physical Science Chapter 2 Review Answers Properties Of Matter Budget-Friendly Options

6. Navigating Physical Science Chapter 2 Review Answers Properties Of Matter eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science Chapter 2 Review Answers Properties Of Matter Compatibility with Devices
 - Physical Science Chapter 2 Review Answers Properties Of Matter Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Chapter 2 Review Answers Properties Of Matter
 - Highlighting and Note-Taking Physical Science Chapter 2 Review Answers Properties Of Matter
 - Interactive Elements Physical Science Chapter 2 Review Answers Properties Of Matter
8. Staying Engaged with Physical Science Chapter 2 Review Answers Properties Of Matter
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Chapter 2 Review Answers Properties Of Matter
9. Balancing eBooks and Physical Books Physical Science Chapter 2 Review Answers Properties Of Matter
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Chapter 2 Review Answers Properties Of Matter
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Chapter 2 Review Answers Properties Of Matter
 - Setting Reading Goals Physical Science Chapter 2 Review Answers Properties Of Matter
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Chapter 2 Review Answers Properties Of Matter
 - Fact-Checking eBook Content of Physical Science Chapter 2 Review Answers Properties Of Matter
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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