

Bonding and Reactions

Study Guide

10 pages including KEY

Chemical Bonding and Reactions Study Guide

This study packet will help you study for the Chemical Bonding and Reactions exam by highlighting the most important concepts on the exam. This test will include both conceptual and short-answer questions on the topics you should know for each section of the material you've studied.

B. VOCABULARY

Atom	Electron	Electron transfer	Electron transfer
Atomic mass	Electron	Electron transfer	Electron transfer
Atomic weight	Electron	Electron transfer	Electron transfer
Atomic	Electron	Electron transfer	Electron transfer
Chemical bond	Electron	Electron transfer	Electron transfer
Chemical reaction	Electron	Electron transfer	Electron transfer

C. READING

1. Why do atoms bond together?

2. What is a covalent bond?

3. What kind of electron flow is covalent bond?

4. What is an ionic bond?

5. What kind of electron flow is ionic bond?

6. Why don't noble gases bond together with other elements?

Why do noble gases not bond together with other elements?

7. Why do noble gases not bond together with other elements?

8. Why do noble gases not bond together with other elements?

Why do noble gases not bond together with other elements?

9. Why do noble gases not bond together with other elements?

10. Why do noble gases not bond together with other elements?

REACTIONS

What happens in a chemical reaction?

Why do some reactions occur in a chemical reaction?

What is a chemical reaction?

What is a chemical reaction?

What is the law of conservation of mass?

What is the law of conservation of mass?

A chemical reaction is a process in which a chemical bond is broken and a chemical bond is formed. The process of breaking a chemical bond is called a chemical reaction. The process of forming a chemical bond is called a chemical reaction.

1. What is a chemical reaction?

Chemical reactions are processes in which a chemical bond is broken and a chemical bond is formed. The process of breaking a chemical bond is called a chemical reaction. The process of forming a chemical bond is called a chemical reaction.

2. What happens in a chemical reaction?

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6. What is a chemical reaction?



B. VOCABULARY

1. Atom: A negatively charged particle that gives matter its mass.

2. Atomic mass: The mass of an atom, measured in atomic mass units (amu).

3. Atomic weight: The average atomic mass of an element, measured in amu.

4. Chemical bond: A force that holds atoms together in a molecule or compound.

5. Chemical reaction: A process in which a chemical bond is broken and a chemical bond is formed.

6. Electron: A negatively charged particle that gives matter its mass.

7. Electron transfer: A process in which an electron is transferred from one atom to another.

8. Electronegativity: A measure of an atom's ability to attract electrons.

9. Electronegativity: A measure of an atom's ability to attract electrons.

10. Electronegativity: A measure of an atom's ability to attract electrons.

Bonding and Chemical Reactions - KEY

C. READING

1. Why do atoms bond together?

Atoms bond together to achieve a more stable electronic configuration, typically by filling or emptying their valence electron shells to satisfy the octet rule.

2. What is a covalent bond?

A covalent bond is a type of chemical bond formed between two atoms by sharing a pair of electrons. The sharing of electrons allows each atom to achieve a more stable electronic configuration.

3. What kind of electron flow is covalent bond?

In a covalent bond, the shared electrons are represented by a pair of dots or a line between the two atoms. The shared electrons are the ones that are shared between the two atoms.

4. What is an ionic bond?

An ionic bond is a type of chemical bond formed between two atoms by the transfer of one or more electrons. The transfer of electrons allows each atom to achieve a more stable electronic configuration.

5. What kind of electron flow is ionic bond?

In an ionic bond, the transferred electrons are represented by a pair of dots or a line between the two atoms. The transferred electrons are the ones that are transferred from one atom to another.

6. Why don't noble gases bond together with other elements?

Noble gases have a full valence shell, which means they have a stable electronic configuration. They do not need to bond with other elements to achieve a more stable electronic configuration.

7. What is a chemical reaction?

A chemical reaction is a process in which a chemical bond is broken and a chemical bond is formed. The process of breaking a chemical bond is called a chemical reaction. The process of forming a chemical bond is called a chemical reaction.

8. What is a chemical reaction?

Chemical reactions are processes in which a chemical bond is broken and a chemical bond is formed.

9. What is a chemical reaction?

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10. What is a chemical reaction?

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11. What is a chemical reaction?

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12. What is a chemical reaction?

Chemical reactions are processes in which a chemical bond is broken and a chemical bond is formed.

13. What is a chemical reaction?

Chemical reactions are processes in which a chemical bond is broken and a chemical bond is formed.

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