

## Density Worksheet

### Physical Science

$D = m/V$

| Substance              | Density (g/cm <sup>3</sup> ) |  | Substance | Density (g/cm <sup>3</sup> ) |
|------------------------|------------------------------|--|-----------|------------------------------|
| Oxygen                 | 0.00133                      |  | Aluminum  | 2.70                         |
| Hydrogen               | 0.000089                     |  | Iron      | 7.87                         |
| Ethanol                | 0.789                        |  | Copper    | 8.96                         |
| Benzene                | 0.880                        |  | Silver    | 10.5                         |
| Water                  | 1.000                        |  | Lead      | 11.34                        |
| Magnesium              | 1.74                         |  | Mercury   | 13.6                         |
| Salt (sodium chloride) | 2.16                         |  | Gold      | 19.32                        |

1. The ratio of an object's mass to its \_\_\_\_\_ is called the density of the object.
2. A kilogram of lead occupies a much smaller volume than a kilogram of water, because \_\_\_\_\_ has a much higher density.
3. For the masses and volumes indicated, calculate the **density** in grams per cubic centimeters.
  - a. mass = 453 g; volume = 225 cm<sup>3</sup>
  - b. mass = 5.0 g; volume = 10.0 cm<sup>3</sup>
  - c. mass = 26.1 g; volume = 2.0 mL
4. If 89.2 mL of a liquid has a mass of 75.2 g, calculate the liquid's density.
5. A cube of metal weighs 1450 g and displaces 542 mL of water when immersed. Calculate the density of the metal.

# Physical Science If8767 Density

**J Spring**



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## **Table of Contents Physical Science If8767 Density**

1. Understanding the eBook Physical Science If8767 Density
  - The Rise of Digital Reading Physical Science If8767 Density
  - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science If8767 Density
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Science If8767 Density
  - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science If8767 Density
  - Personalized Recommendations
  - Physical Science If8767 Density User Reviews and Ratings
  - Physical Science If8767 Density and Bestseller Lists

5. Accessing Physical Science If8767 Density Free and Paid eBooks
  - Physical Science If8767 Density Public Domain eBooks
  - Physical Science If8767 Density eBook Subscription Services
  - Physical Science If8767 Density Budget-Friendly Options
6. Navigating Physical Science If8767 Density eBook Formats
  - ePub, PDF, MOBI, and More
  - Physical Science If8767 Density Compatibility with Devices
  - Physical Science If8767 Density Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Physical Science If8767 Density
  - Highlighting and Note-Taking Physical Science If8767 Density
  - Interactive Elements Physical Science If8767 Density
8. Staying Engaged with Physical Science If8767 Density
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physical Science If8767 Density
9. Balancing eBooks and Physical Books Physical Science If8767 Density
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical Science If8767 Density
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Physical Science If8767 Density
  - Setting Reading Goals Physical Science If8767 Density
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science If8767 Density
  - Fact-Checking eBook Content of Physical Science If8767 Density
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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