

2-2

Possible stereoisomers are:

a) **Eis:** **Tirs**, random (because of double bond)  
 b) **Isotactic**, **syndiotactic**, **atactic** (because of  $-\text{CHCl}-$  groups). However, because of distance between  $-\text{CHCl}-$  groups tacticity is unlikely to influence ability to crystallize unless rest of groups are all gls or trans.

Use  $\left[ \frac{1}{\sin \theta} \right]^2$  for  $\text{C-O} = a$   $108^\circ$  for  $\text{C-O-C} = \theta$   
 $\text{O-C} = b$   $110^\circ$  for  $\text{C-C-O} = \theta$

2-3

Graphical solution is easiest using protractor and ruler. Vector summation can be used, also.



If by syndiotactic we mean a succession of **d** and **l** monomer units, the distance turns out to be the same. Another structure is possible with methyls alternating so that the repeat distance would be doubled.

2-4

Refer to Table 2-1, Fig. 2-1.  
 Polycyclohexane in cyclohexane:

$$X = 0.136, \quad V_u = 108, \quad v_2 = 0.10$$

$$NV = 2.45 \times 10^3, \quad \text{so that } N = 2.27 \times 10^5 \text{ units/cm}^3$$

Same in toluene:

$$X = 0.557, \quad V_u = 114, \quad W_u = 2.30 \times 10^{-22}, \quad \text{then } v_2 = 0.21$$

$\frac{10^{-22} \text{ moles}}{1 \text{ cc}} \div \frac{1}{2} = \frac{1}{2} \text{ original volume.}$

2-5

$$N = 0.5, \quad \text{moles/volume} = (\text{g/vol})/(\text{g/mole})$$

If  $M = 161.91$ , per repeat unit = 58 for  $\text{C}_5\text{H}_8\text{O}$  (C chain atoms)  
 then  $M_c = 161.91$ , between cross-links =  $\frac{5800 \times 58}{5} \text{ g/mole}$

$$\rho = 0.913, \quad a = 10^{-8} \text{ meter}$$

$$V = 102 \times 1.21 \times 10^{-22} / 0.800 = 1.58 \times 10^{-22}$$

From Fig. 2-1:  $v_2 = 0.059$

same additive volumes. Basis: Total volume = 1 cc

$$1.0 \text{ cc} = V + V' + V_2 = 0.931 + 0.069$$

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