

Self Assessment Paper-3

Science

Class- X, Session: 2022-23

UNSOLVED

Time Allowed: 3 hours

Maximum Marks: 40

General Instructions:

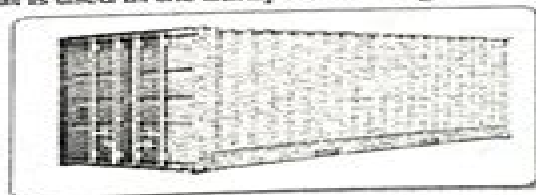
- This question paper consists of 39 questions in 5 sections.
- All questions are compulsory. However, an internal choice is provided in some questions. A student is expected to attempt only one of these questions.
- Section A consists of 20 objective type questions carrying 1 mark each.
- Section B consists of 6 Very Short questions carrying 02 marks each. Answers to these questions should be in the range of 30 to 50 words.
- Section C consists of 7 Short Answer type questions carrying 03 marks each. Answers to these questions should be in the range of 50 to 80 words.
- Section D consists of 3 Long Answer type questions carrying 05 marks each. Answer to these questions should be in the range of 80 to 120 words.
- Section E consists of 3 source-based/case-based units of assessment of 04 marks each with sub-parts.

Section-A

20 Marks

(Select and write one most appropriate option out of the four options given for each of the questions 1 – 20)

- Three beakers labelled as A, B and C each containing 25 mL of water was taken. A small amount of NaOH, anhydrous CuSO_4 and NaCl were added to the beakers A, B and C respectively. It was observed that there was an increase in the temperature of the solutions contained in beakers A and B, whereas in case of beaker C, the temperature of the solution falls. Which one of the following statement(s) is (are) correct?
(i) In beakers A and B, exothermic process has occurred.
(ii) In beakers A and B, endothermic process has occurred.
(iii) In beaker C, exothermic process has occurred.
(iv) In beaker C, endothermic process has occurred.
(A) (i) only
(B) (ii) only
(C) (i) and (iv)
(D) (iii) and (iii)
OR
(B) 7 covalent bonds
(D) 9 covalent bonds
- Ethane, with the molecular formula C_2H_6 has:
(A) 6 covalent bonds
(C) 8 covalent bonds
- Shown below is a container that is used in the transportation of goods over long distances.



These containers are made of steel. Which property of steel is mainly used to make these containers?

- (A) Its ductility
(B) Its malleability
(C) Its metallic lustre
(D) Its electrical conductivity
- In the given chemical reactions, Salt X \rightarrow Salt Y + Gas Z.
 - Salt X is commonly used in bakery product.
 - Salt Y is common constituent of washing powder.
 - Gas Z turns lime water milky.

Sample Paper Unsolved

Thomas Griffiths



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