

CHEMISTRY 101 Practice Exam 1 DR. PECK

- Directions:
- (1) Choose the best answer for each multiple choice question (numbers 1 - 15). Transfer your multiple choice answers onto the scantron.
 - (2) Do not separate any of the pages. When finished, give the proctor at the front of the room both the exam and the completed scantron. Cover your exam and your work whenever possible.

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(NOTE: This practice exam does not contain questions based upon topics from Chapter 5. The actual exam may or may not depending upon our ability to cover Chapter 5 in lecture before the date of the first exam.)

1. Which one of the following is a weak acid?
 (a) HBr (b) HClO₄ (c) HNO₃ (d) HCl (e) HClO₂

2. What is the oxidizing agent in the following reaction?

$$6\text{KOH(aq)} + 3\text{Cl}_2\text{(g)} \rightarrow \text{KClO}_3\text{(aq)} + 5\text{KCl(aq)} + 3\text{H}_2\text{O(l)}$$
 (a) KOH (b) KCl (c) KClO₃ (d) Cl₂ (e) H₂O

3. Balance the following equation with the **smallest whole number coefficients**. What is the coefficient for HCl in the balanced equation?

$$\text{SnS}_2 + \text{HCl} \rightarrow \text{H}_2\text{SnCl}_6 + \text{H}_2\text{S}$$
 (a) 4 (b) 6 (c) 3 (d) 12 (e) 2

4. Classify the following reaction by giving **all** of these reaction type(s) that apply.

I. redox	II. combination	III. decomposition
IV. displacement	V. metathesis	

$$2\text{PbO(s)} + \text{O}_2\text{(g)} \rightarrow 2\text{PbO}_2\text{(s)}$$
 (a) I and V (b) only II (c) I and II (d) only III (e) only V

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Annelies Wilder-Smith



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Table of Contents Sample Questions On Chemistry 101

1. Understanding the eBook Sample Questions On Chemistry 101
 - The Rise of Digital Reading Sample Questions On Chemistry 101
 - Advantages of eBooks Over Traditional Books
2. Identifying Sample Questions On Chemistry 101
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Sample Questions On Chemistry 101
 - User-Friendly Interface
4. Exploring eBook Recommendations from Sample Questions On Chemistry 101
 - Personalized Recommendations
 - Sample Questions On Chemistry 101 User Reviews and Ratings
 - Sample Questions On Chemistry 101 and Bestseller Lists

5. Accessing Sample Questions On Chemistry 101 Free and Paid eBooks
 - Sample Questions On Chemistry 101 Public Domain eBooks
 - Sample Questions On Chemistry 101 eBook Subscription Services
 - Sample Questions On Chemistry 101 Budget-Friendly Options
6. Navigating Sample Questions On Chemistry 101 eBook Formats
 - ePub, PDF, MOBI, and More
 - Sample Questions On Chemistry 101 Compatibility with Devices
 - Sample Questions On Chemistry 101 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Sample Questions On Chemistry 101
 - Highlighting and Note-Taking Sample Questions On Chemistry 101
 - Interactive Elements Sample Questions On Chemistry 101
8. Staying Engaged with Sample Questions On Chemistry 101
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Sample Questions On Chemistry 101
9. Balancing eBooks and Physical Books Sample Questions On Chemistry 101
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Sample Questions On Chemistry 101
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Sample Questions On Chemistry 101
 - Setting Reading Goals Sample Questions On Chemistry 101
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Sample Questions On Chemistry 101
 - Fact-Checking eBook Content of Sample Questions On Chemistry 101
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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