

CBSE Sample Paper Class 7 Science SA1 Set 1

SUBJECT: SCIENCE Class - VII

TIME: 21/2Hrs. MAX. MARKS: 60

General Instructions

1) The question paper comprises of two sections, A and B. You are to attempt both the sections. All questions of section A and B are to be attempted separately.

2) All questions are compulsory.

3) Marks for each question are indicated against it.

Choose the most appropriate option: -

CENTRAL A

	SECTION: - A	
1.	Name any one insectivorous plant.	1
2.	Name the carbohydrate which is the instant source of energy.	1
3.	Name any two natural indicator.	1
4.	What type of climate is there in Rajasthan?	1
5.	By which name a cyclone is known in Japan.	1. 3
6.	Distinguish between a parasite and a saprotroph.	2
7.	Give two examples each of conductor and insulator.	2
8.	Why an antacid tablet is taken when you suffer from acidity.	2
9.	What do you mean by reeling of silk?	2
10	. Why plants are called autotrophs.	2
1.1	. Why wearing more layers of clothing during winter keep us warmer	
han wearing just one thick piece of clothing?		
12	.Explain how painting of an iron gate prevent it from rusting.	3
1.3	.Draw a diagram showing different stages of life history of silk moth.	3
14	.State differences between acid and bases.	3
1.5	.Classify the changes involved in the following process as physical or	chemical
	change 3	
	(a) Photosynthesis	
	(b) Dissolving sugar in water.	
	(c) Burning of coal.	
	(d) Melting of wax.	
	(e) Beating of aluminum to make aluminum foil.	
	(f) Digestion of food.	
16	Draw a well labeled diagram to show digestive system in humans.	5
17	.Explain the adaptation of polar bear to live in polar region.	5

Sample Paper For Class Science Sa1

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Sample Paper For Class Science Sa1:

MTG ScoreMore CBSE Sample Papers for Class 10 Science (For 2024 Exams) MTG Editorial Board, 2023-10-10 MTG has released the latest updated CBSE ScoreMore 15 Sample Papers for the CBSE Session 2023 2024 of Class 10 It comprises a set of 15 sample papers 1 official CBSE sample question paper and 14 sample questions papers according to the blueprint issued by CBSE on 31st March 2023 Each question paper covers subjective and objective type questions including all question typologies Very Short Answer Short Answer Long Answer Type Questions Case Based and MCQs Assertion Reason Self evaluation sheet is included to check your readiness and performance Detailed solutions of each sample question are also provided in the book CBSE Additional Practice Questions released on 8th September are also given in the book Online support in the form of CBSE Tips Exam Alerts Formula Book Learn Retain Chapter wise Objective Questions and 2023 CBSE Solved Paper can also be availed with this book Educart CBSE Science Class 10 Sample Papers 2024-25 (With exclusive CBSE Mock Booklets for 2025 Exam) Educart, 2024-09-17 What You Get Chapter wise Revision Maps 3 Most Likely Sets with Answer Booklets Educart CBSE Science Class 10 Sample Papers 2024 25 With exclusive CBSE Mock Booklets for 2025 Exam Based on the CBSE Class 10 syllabus for 2025 exams Chapter wise revision maps for better concept clarity Competency based questions are included as per the new exam pattern Tricks and keywords for better time management Stepwise marks breakdown for understanding the strengths and weaknesses Practice the 3 most likely guestion sets with their answer booklets Why choose this book Guarantee the complete score in 40% of the board examination paper and increase your chances to become the topper 10 Sample Question Papers for CBSE Class 10 Science with Marking Scheme & Revision Notes Disha Experts, The book provides 10 Sample Question Papers for CBSE Class 10 Science March 2018 Exam designed exactly as per the latest Blue Prints and Sample Papers issued by CBSE Each of the Sample Paper provides detailed solutions with Marking Scheme Further the book provides 1 CBSE Sample Paper with Solutions CBSE Blueprint issued by the CBSE Board The book also provides Revision Notes which will help you in revising the syllabus quickly before the exam The book is made strictly in accordance with the latest CBSE prescribed syllabus and pattern 50 Sample Papers for CBSE Class 10 Science, Mathematics, Social Science, Hindi B and English Language & Literature 2020 Exam Disha Experts, 2019-10-21

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web elements in the second row of the periodic table place their electrons in the 2n shell as well as the 1n shell for instance lithium li has three electrons two fill the 1 s orbital and the third is placed in the 2 s orbital giving an electron configuration of 1 s 2 2 s 1

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web electron configurations and trends answer these discussion questions given the following part of an electron configuration 3s2 explain which part refers to the number of electrons the energy level and the sublevel what is the maximum number of electrons that can occupy a single orbital state and explain hund s rule

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filled to their maximum bracketed noble gas symbols on the left represent inner configurations that are the same in each period written out these are he 2 helium $1s\ 2$ ne 10 neon $1s\ 2\ 2s\ 2$ $2p\ 6$

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