# NCERT Exemplar for Class 7 Science Chapter 6 Physical and Chemical Changes

4. Which of the following is/are true when milk changes into curd?

- (i) Its state is changed from liquid to semi solid.
- (iii) It changes colour.
- (iii) It changes taste.
- (iv) The change cannot be reversed. Choose the correct option from below :

(a) (i) and (ii) are correct

- (b) (ii) and (iii) are correct
- (c) (i), (iii) and (iv) are correct
- (d) (i) to (iv) are correct

#### Soln:

Answer is (c) (i), (iii) and (iv) are correct

#### Solution:

When milk turn to turn it changes its shape to semisolid, its taste turn sour but its color remains white hence answer is c)

5. A man painted his main gate made up of iron, to

- (i) prevent it from rusting.
- (iii) protect it from sun.
- (iii) make it look beautiful.
- (iv) make it dust free.

Which of the above statement(s) is/are correct?

- (a) (ii) and (iii)
- (b) (ii) and (iii)
- (c) only (ii)
- (d) (i) and (iii)

#### Soln:

Answer is (d) (i) and (iii)

6. Iron pillar near the Qutub Minar in Delhi is famous for the following facts. Which of these facts is responsible for its long stability?

- (a) It is more than 7 metres high.
- (b) It weighs about 6000 kg.
- (c) It was built more than 1600 years ago.
- (d) It has not rusted after such a long period.

#### Soln:

Answer is (d) It has not rusted after such a long period.

# **Pearson Chemical Changes Answer Key Ch 6**

F. Albert Cotton, Geoffrey Wilkinson

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The Biological Chemistry of the Elements J. J. R. Frausto da Silva, Robert Joseph Paton Williams, 2001-08-16 This text describes the functional role of the twenty inorganic elements essential to life in living organisms Focus on Physical Science California Edition Michael J. Padilla, 2007 Scott Foresman Science ,2008 Stereochemistry and the **Chemistry of Natural Products** Ivor Lionel Finar, 1956 The Aqueous Chemistry of Oxides Bruce C. Bunker, William H. Casey, 2016-02-02 The Aqueous Chemistry of Oxides is a single volume text that encapsulates all of the critical issues associated with how oxide materials interact with aqueous solutions It serves as a central reference for academics working with oxides in the contexts of geology various types of inorganic chemistry and materials science. The text also has utility for professionals working with industrial applications in which oxides are either prepared or must perform in aqueous environments The volume is organized into five key sections Part One features two introductory chapters intended to introduce the mutual interests of engineers chemists geologists and industrial scientists in the physical and chemical properties of oxide materials Part Two provides the essential and fundamental principles that are critical to understanding most of the major reactions between water and oxides Part Three deals with the synthesis of oxide materials in aqueous media Part Four deals with oxide water reactions and their environmental and technological impacts and Part Five is devoted to other types of relevant reactions The Aqueous Chemistry of Oxides is the first book that provides a comprehensive summary of all of the critical reactions between oxides and water in a single volume As such it ties together a wide range of existing books and literature into a central location that provides a key reference for understanding and accessing a broad range of more specialized topics The book contain over 300 figures and tables Organic Chemistry: Stereochemistry and the chemistry of natural products Ivor Lionel Finar,1956 Physical Chemistry Henry Eyring, Douglas Henderson, Wilhelm Advanced Inorganic Chemistry F. Albert Cotton, Geoffrey Wilkinson, 1988-04-05 New York Wiley c1988 Jost, 1967

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A Textbook of Physical Chemistry Arthur Adamson, 2012-12-02 A Textbook of Physical Chemistry Second Edition serves as an introductory text to physical chemistry Topics covered range from wave mechanics and chemical bonding to molecular spectroscopy and photochemistry ideal and nonideal gases the three laws of thermodynamics thermochemistry and solutions of nonelectrolytes The kinetics of gas phase reactions colloids and macromolecules and nuclear chemistry and radiochemistry are also discussed This edition is comprised of 22 chapters the first of which introduces the reader to the behavior of ideal and nonideal gases with particular emphasis on the van der Waals equation The discussion then turns to the kinetic

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