

Place Value Study Guide

Test Date: _____

Digits: 0, 1, 2, 3, 4, 5, 6, 7, 8, 9

Place Value Names and Locations:

Hundred Thousands	Ten Thousands	Thousands	Hundreds	Tens	Ones
3	1	6	7	5	8



The comma is located in between the hundreds place and thousands place.

Place: The **LOCATION** of a digit in a number

- 316,758 - The 8 is in the **ones** place.
- 316,758 - The 5 is in the **tens** place.
- 316,758 - The 7 is in the **hundreds** place.
- 316,758 - The 6 is in the **thousands** place.
- 316,758 - The 1 is in the **ten thousands** place.
- 316,758 - The 3 is in the **hundred thousands** place.

Value: How much a digit is **WORTH**.

- 316,758 - The value is 8 because the 8 is in the **ones** place.
- 316,758 - The value is 50 because the 5 is in the **tens** place.
- 316,758 - The value is 700 because the 7 is in the **hundreds** place.
- 316,758 - The value is 6,000 because the 6 is in the **thousands** place.
- 316,758 - The value is 10,000 because the 1 is in the **ten thousands** place.
- 316,758 - The value is 300,000 because the 3 is in the **hundred thousands** place.

Place Value Study Guide

M Carnoy



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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper's mass into account add these to the beaker 3 measure 50 0 ml of water in the

cylinder and add it to the beaker 4

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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web under the finalize of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away cooper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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