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Paper - 3

(2nd Semester Examination 2023-24)CHEMISTRY

1. A student investigates the rate of reaction between magnesium ribbon and excess dilute hydrochloric acid by measuring the volume of gas produced. The student uses the apparatus in Fig. 1.1 to do two different experiments.

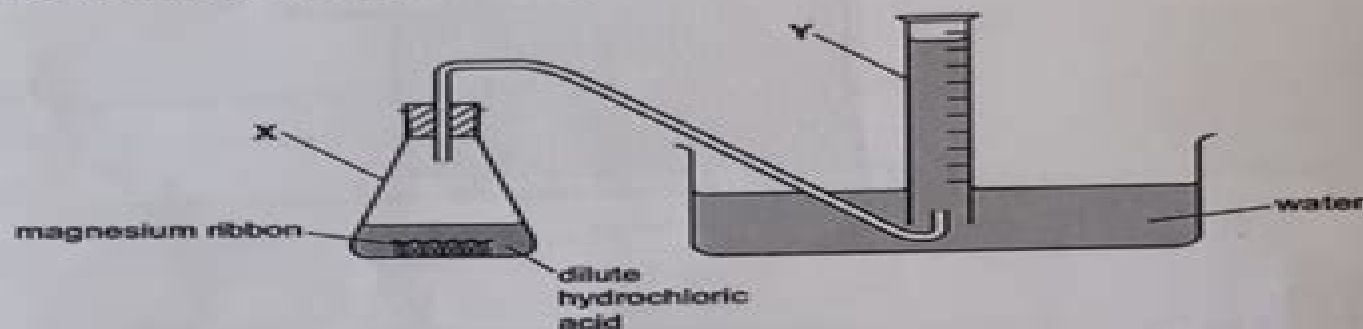


Fig. 1.1

- (a) Name the items of apparatus labelled X and Y in Fig. 1.1 [2]
- X Round bottom conical flask
- Y measuring cylinder
- (b) The gas made in the reaction is hydrogen. [1]

Describe how the student can test that the gas is hydrogen.

Give the expected result of the test.

test put a burning splint near the beaker & if it creates a popping sound it is hydrogen

result Bubbles will form & it will create a popping sound, by the burning splint.

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