

REVISION: Global and International "Text" in Chemistry in the Environment (Exercises)

Name:	Matr.:	Grp.:	Grp.:	Grp.:
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Write the answers in your own words! Some of the most important chemical reactions have been significantly changed!

Explain the following: Explain what the effect of the following

change is: change of concentration or change of pressure or temp.



Part 1: Knowledge and Understanding (20 marks)

Question 1 (4 marks)

What is a chemical reaction? In the open system, equilibrium is reached when the concentration of

- _____ is constant or has reached an
equilibrium state.
Equilibrium of two independent
Equilibrium
- _____ is chemical equilibrium is reached experimentally, which is the state of equilibrium. The state of equilibrium is reached when the
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- _____ When a system is at equilibrium, the concentration of
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What are the units of the rate?

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Sch4c Environmental Chemistry Unit Test

Jacquelyn A. Mitchell



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