

Three-mechanism reactions

10. A reaction $\text{A} \rightarrow \text{B}$ has $E_a = 100 \text{ kJ mol}^{-1}$. At 25°C , $k = 1.0 \times 10^{-5} \text{ s}^{-1}$. At 35°C , $k = 1.0 \times 10^{-4} \text{ s}^{-1}$.

Which statement is correct for this reaction?

- a. A and B are approximately equal in energy.
- b. A and B are approximately equal in E_a .
- c. At 35°C , the forward reaction is faster.
- d. A and B are approximately equal in E_a .
- e. A and B are approximately equal in energy.

11. The energy profile diagram for the reaction $\text{A} \rightarrow \text{B}$ is shown below. The energy of A is 100 kJ mol^{-1} .

What is the energy of B ?

12. The reaction $\text{A} \rightarrow \text{B}$ is exothermic. The energy profile diagram for the reaction is shown below.

What is the energy of B ?

13. The energy profile diagram for the reaction $\text{A} \rightarrow \text{B}$ is shown below. The energy of A is 100 kJ mol^{-1} .

14. The reaction $\text{A} \rightarrow \text{B}$ is exothermic. The energy profile diagram for the reaction is shown below.

15. The reaction $\text{A} \rightarrow \text{B}$ is exothermic. The energy profile diagram for the reaction is shown below.

16. The reaction $\text{A} \rightarrow \text{B}$ is exothermic. The energy profile diagram for the reaction is shown below.

17. The reaction $\text{A} \rightarrow \text{B}$ is exothermic. The energy profile diagram for the reaction is shown below.

- a. 100 kJ mol^{-1}
- b. 150 kJ mol^{-1}
- c. 200 kJ mol^{-1}
- d. 250 kJ mol^{-1}

18. The reaction $\text{A} \rightarrow \text{B}$ is exothermic. The energy profile diagram for the reaction is shown below.

19. The reaction $\text{A} \rightarrow \text{B}$ is exothermic. The energy profile diagram for the reaction is shown below.

20. The reaction $\text{A} \rightarrow \text{B}$ is exothermic. The energy profile diagram for the reaction is shown below.

- a. 100 kJ mol^{-1}
- b. 150 kJ mol^{-1}
- c. 200 kJ mol^{-1}
- d. 250 kJ mol^{-1}

21. The reaction $\text{A} \rightarrow \text{B}$ is exothermic. The energy profile diagram for the reaction is shown below.

22. The reaction $\text{A} \rightarrow \text{B}$ is exothermic. The energy profile diagram for the reaction is shown below.

23. The reaction $\text{A} \rightarrow \text{B}$ is exothermic. The energy profile diagram for the reaction is shown below.

24. The reaction $\text{A} \rightarrow \text{B}$ is exothermic. The energy profile diagram for the reaction is shown below.

25. The reaction $\text{A} \rightarrow \text{B}$ is exothermic. The energy profile diagram for the reaction is shown below.

26. The reaction $\text{A} \rightarrow \text{B}$ is exothermic. The energy profile diagram for the reaction is shown below.

27. The reaction $\text{A} \rightarrow \text{B}$ is exothermic. The energy profile diagram for the reaction is shown below.

28. The reaction $\text{A} \rightarrow \text{B}$ is exothermic. The energy profile diagram for the reaction is shown below.

29. The reaction $\text{A} \rightarrow \text{B}$ is exothermic. The energy profile diagram for the reaction is shown below.

30. The reaction $\text{A} \rightarrow \text{B}$ is exothermic. The energy profile diagram for the reaction is shown below.

- a. 100 kJ mol^{-1}
- b. 150 kJ mol^{-1}
- c. 200 kJ mol^{-1}
- d. 250 kJ mol^{-1}

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Rachel S Tattersall



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