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#### CHAPTER 11 REVIEW

## Molecular Composition of Gases

#### MIXED REVIEW

SHORT ANSWER Answer the following questions in the space provided.

- 1. \_\_\_\_\_ The average speed of a gas molecule is most directly related to the \_\_\_\_
  - (a) polarity of the molecule
  - (b) pressure of the gas
  - (c) temperature of the gas
  - (d) number of moles in the sample
- Which of the following statements best represents Graham's law for two gases at the same temperature?
  - (a) The molar masses and the velocities are inversely proportional.
  - (b) The molar masses and the velocities are directly proportional.
  - (c) The molar masses and the square roots of the velocities are directly proportional.
  - (d) The molar masses and the squares of the velocities are inversely proportional.
- 3. Water is synthesized from its elements according to the following equation:

$$2H_2(g) + O_2(g) \rightarrow 2H_2O(l)$$

- a. Which is greater, the volume of H<sub>2</sub> consumed or the volume of O<sub>2</sub> consumed?
- b. Which is greater, the mass of H<sub>2</sub> consumed or the mass of O<sub>2</sub> consumed?
- 4. Consider a 2.24 L sample of an ideal gas at STP.
  - \_\_\_\_\_ a. How many moles are present in the sample, or is more data needed?
  - b. How many molecules are present in the sample, or is more data needed?
  - e. What is the mass of the sample, or is more data needed?

PROBLEMS Write the answer on the line to the left. Show all your work in the space provided.

Calculate the volume of 3.5 mol of an ideal gas at 24°C and 760 torr.

# **Review Molecular Composition Of Gases Section 2**

**J Dewey** 

#### **Review Molecular Composition Of Gases Section 2:**

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#### **Table of Contents Review Molecular Composition Of Gases Section 2**

- 1. Understanding the eBook Review Molecular Composition Of Gases Section 2
  - The Rise of Digital Reading Review Molecular Composition Of Gases Section 2
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Review Molecular Composition Of Gases Section 2
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Review Molecular Composition Of Gases Section 2
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Review Molecular Composition Of Gases Section 2
  - Personalized Recommendations
  - Review Molecular Composition Of Gases Section 2 User Reviews and Ratings
  - Review Molecular Composition Of Gases Section 2 and Bestseller Lists
- 5. Accessing Review Molecular Composition Of Gases Section 2 Free and Paid eBooks
  - Review Molecular Composition Of Gases Section 2 Public Domain eBooks
  - Review Molecular Composition Of Gases Section 2 eBook Subscription Services
  - Review Molecular Composition Of Gases Section 2 Budget-Friendly Options

- 6. Navigating Review Molecular Composition Of Gases Section 2 eBook Formats
  - o ePub, PDF, MOBI, and More
  - Review Molecular Composition Of Gases Section 2 Compatibility with Devices
  - Review Molecular Composition Of Gases Section 2 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - o Adjustable Fonts and Text Sizes of Review Molecular Composition Of Gases Section 2
  - Highlighting and Note-Taking Review Molecular Composition Of Gases Section 2
  - Interactive Elements Review Molecular Composition Of Gases Section 2
- 8. Staying Engaged with Review Molecular Composition Of Gases Section 2
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Review Molecular Composition Of Gases Section 2
- 9. Balancing eBooks and Physical Books Review Molecular Composition Of Gases Section 2
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Review Molecular Composition Of Gases Section 2
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Review Molecular Composition Of Gases Section 2
  - Setting Reading Goals Review Molecular Composition Of Gases Section 2
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Review Molecular Composition Of Gases Section 2
  - Fact-Checking eBook Content of Review Molecular Composition Of Gases Section 2
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements

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